

# Physical and chemical properties of the Upper Muschelkalk aquifer in Northern Switzerland

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& L. Diamond

Acknowledgments:

Nagra

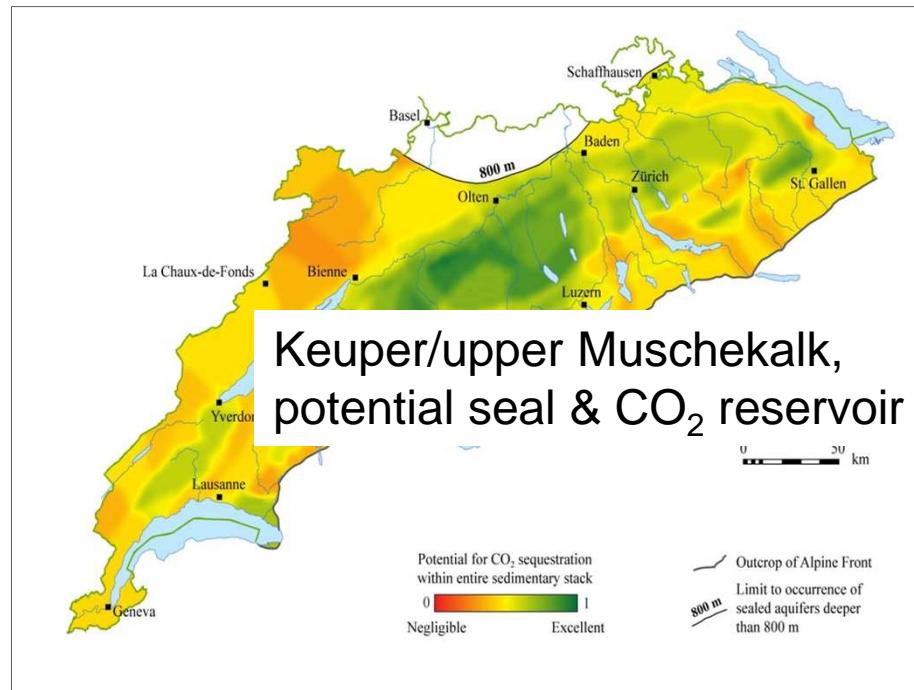
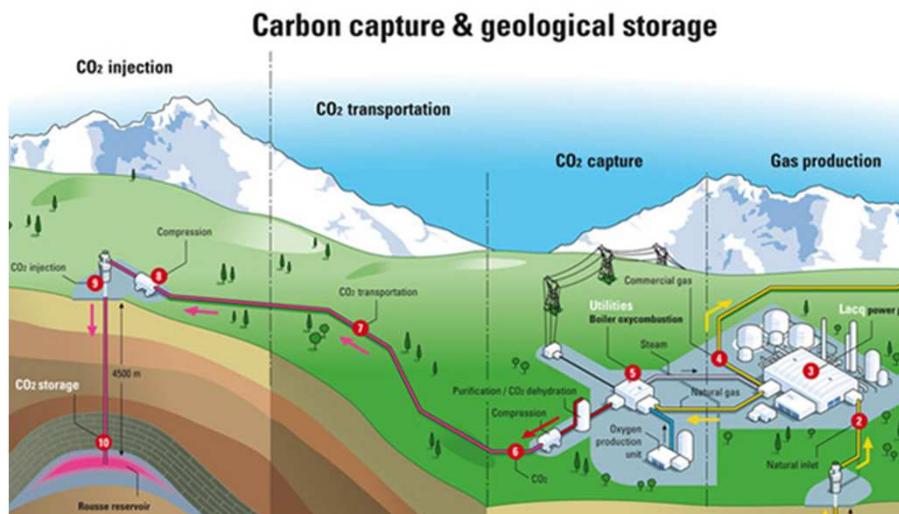
Ronny Pini (Stanford U.)

Dorian Marx (ETHZ)

Carbon management in energy  
generation (CARMA)



# Motivation



Chevalier et al. (2010) Swiss Journal of Geosciences, Vol. 203

**Where is the unit porous and permeable?  
(spatial distribution in the Swiss Molasse Basin)**

**How did the porosity and permeability form?  
(sedimentary, diagenetic and tectonic history)**



Study area



Switzerland



Benken (822m)



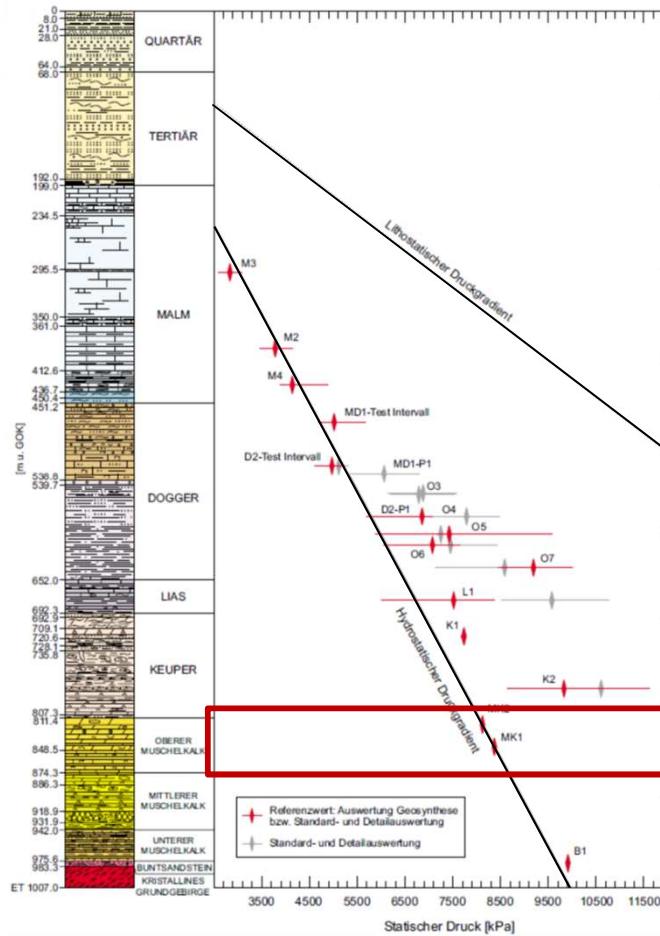
Weiach (829m)



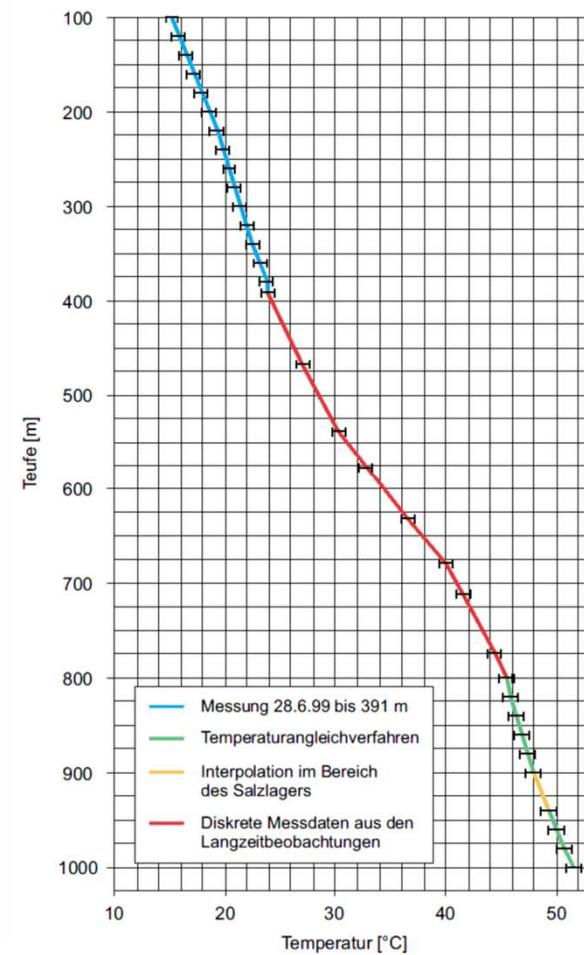
Mineral composition:  
Dolomite  
Calcite  
Anhydrite  
Quartz  
Kaolinite  
Iron sulphides

## Pressure and temperature effects (Benken)

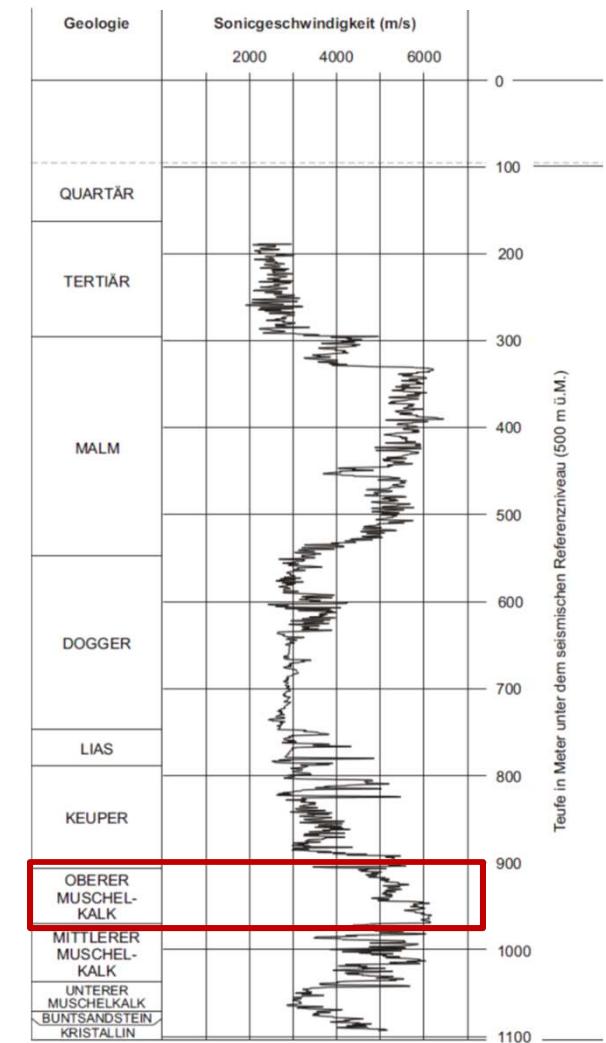
### Pore pressure



### Geothermal gradient

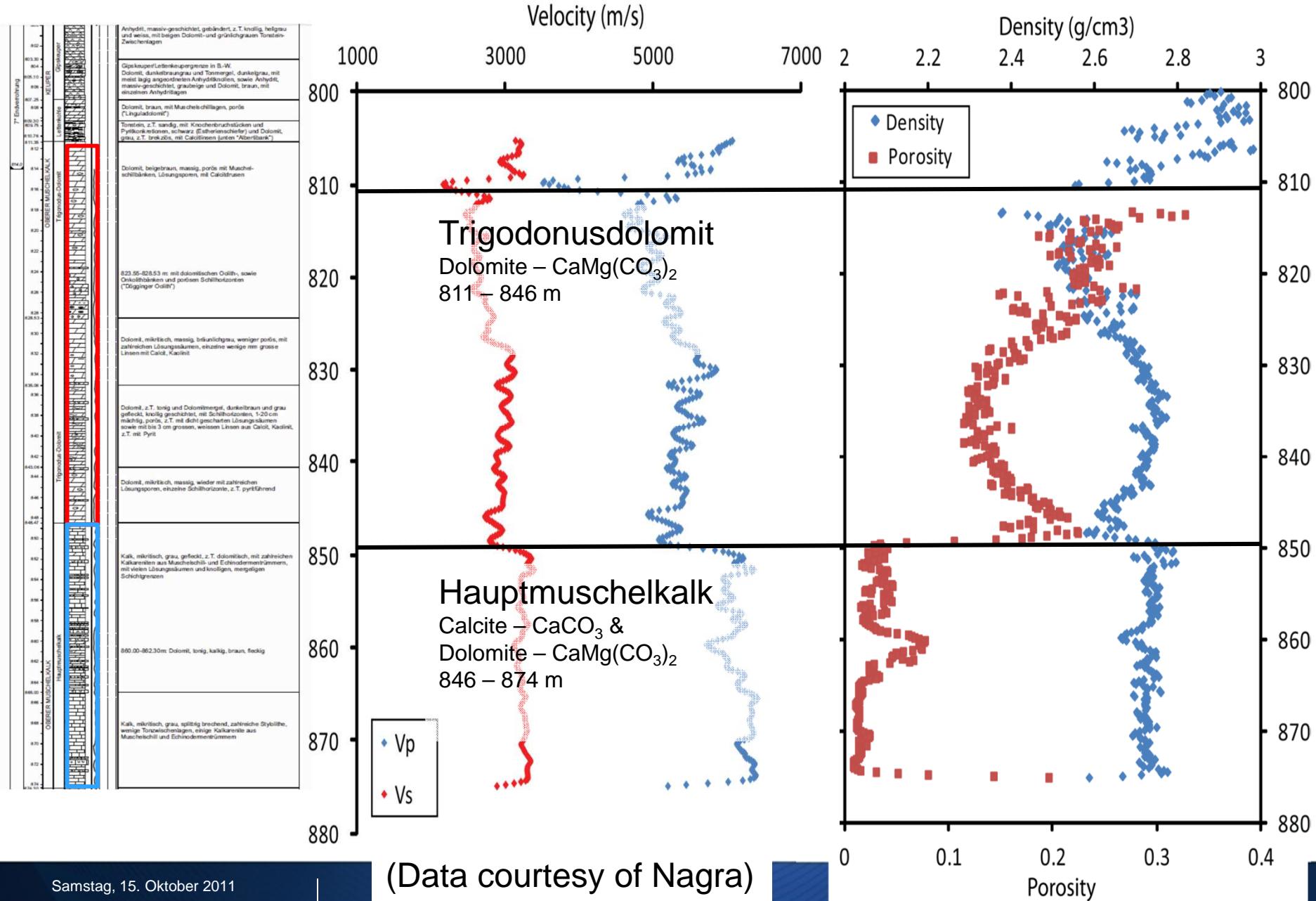


### Sonic velocity (m/s)



(Nagra, 1999)

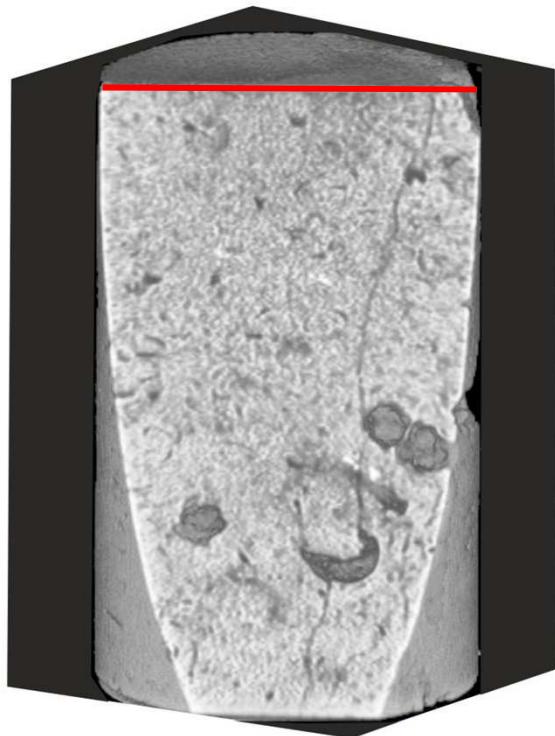
# Upper Muschelkalk borehole logging record



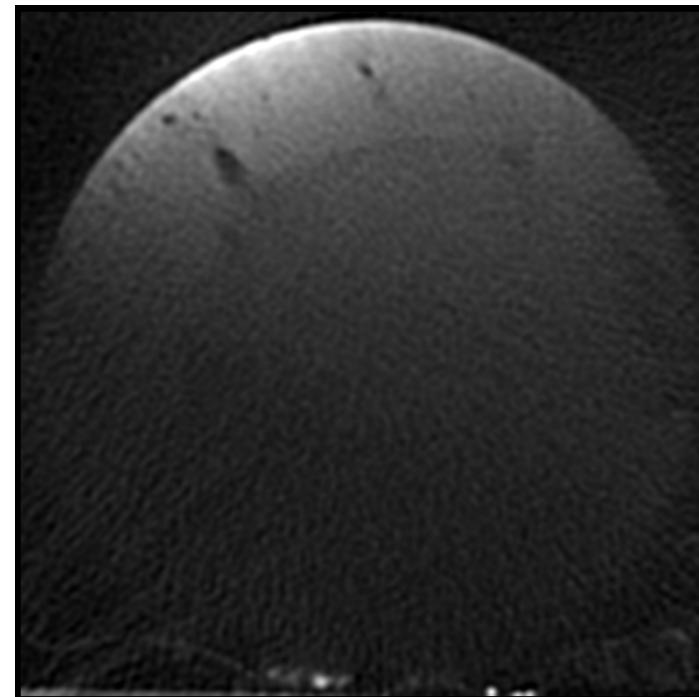
# Macroporosity – hospital CT scan

827 mbs

Towards surface

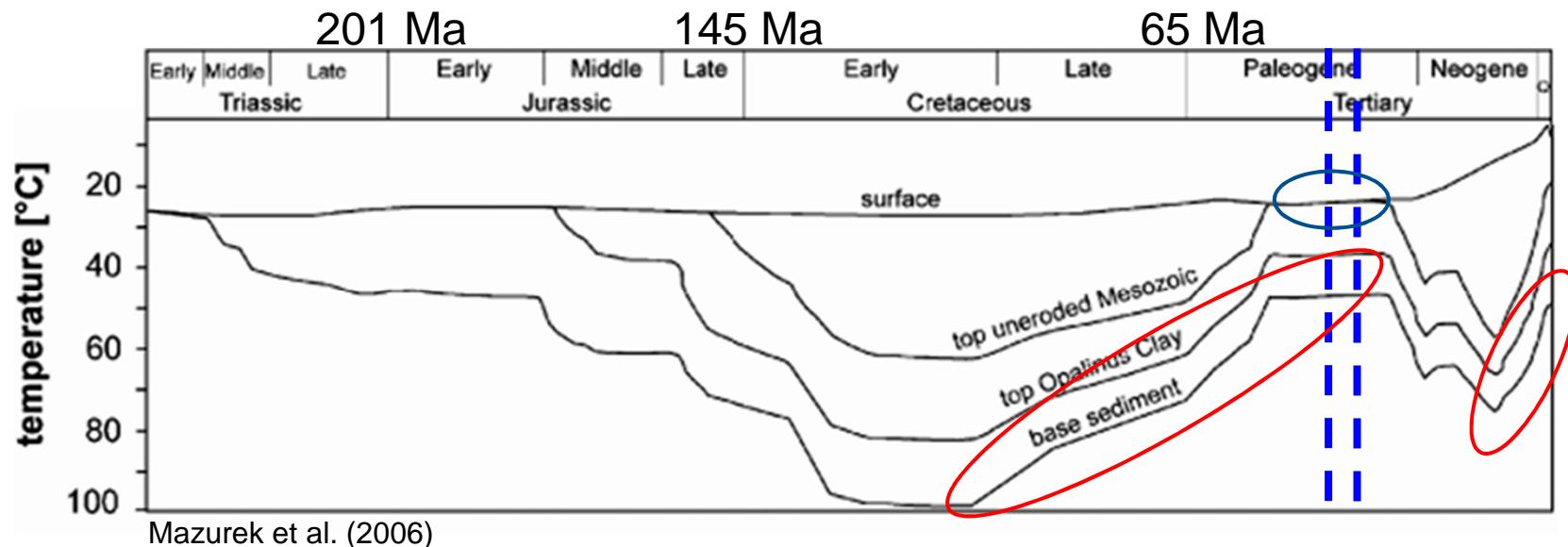


10 cm



Resolution = 0.6 mm

# Origin of the macroporosity



Fluid inclusion data indicates that quartz grows during uplift of basin

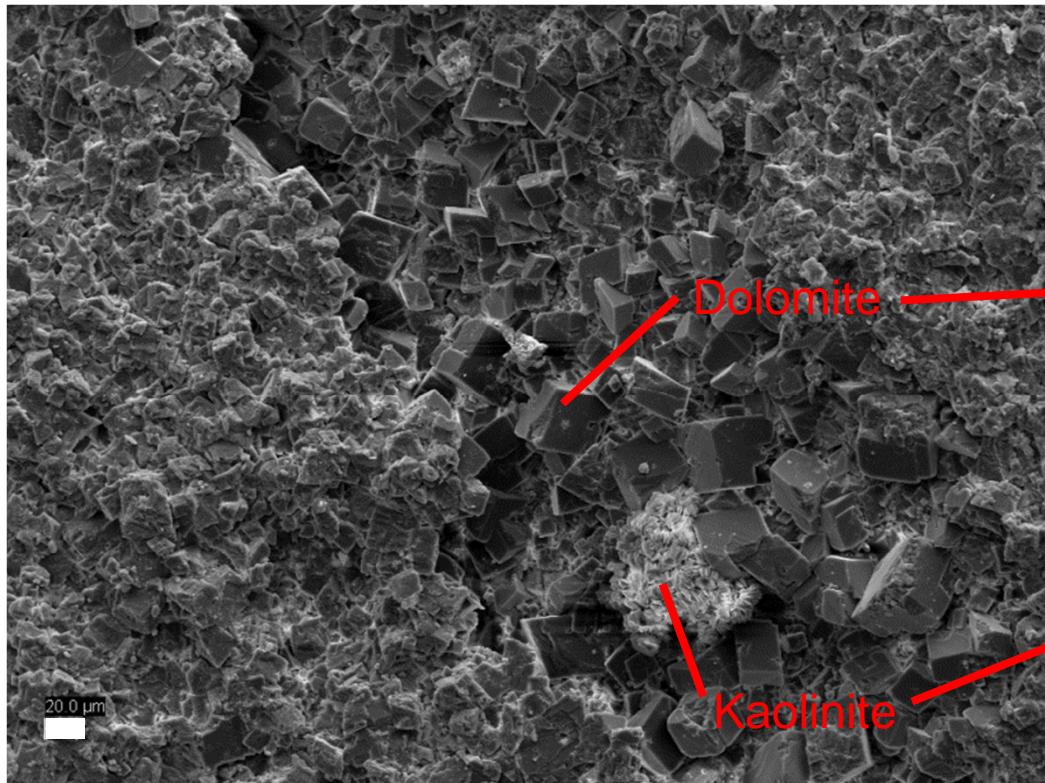
Oxygen isotope data of quartz indicates influence of meteoric water during crystal growth  
→ infiltration Eocene/Oligocene

Signal from Eocene/Oligocene boundary via hydrogen isotope data of kaolinite

**Timing of dissolution of anhydrite nodules and creation of macroporosity most likely during late stage uplift in the Upper Cretaceous**

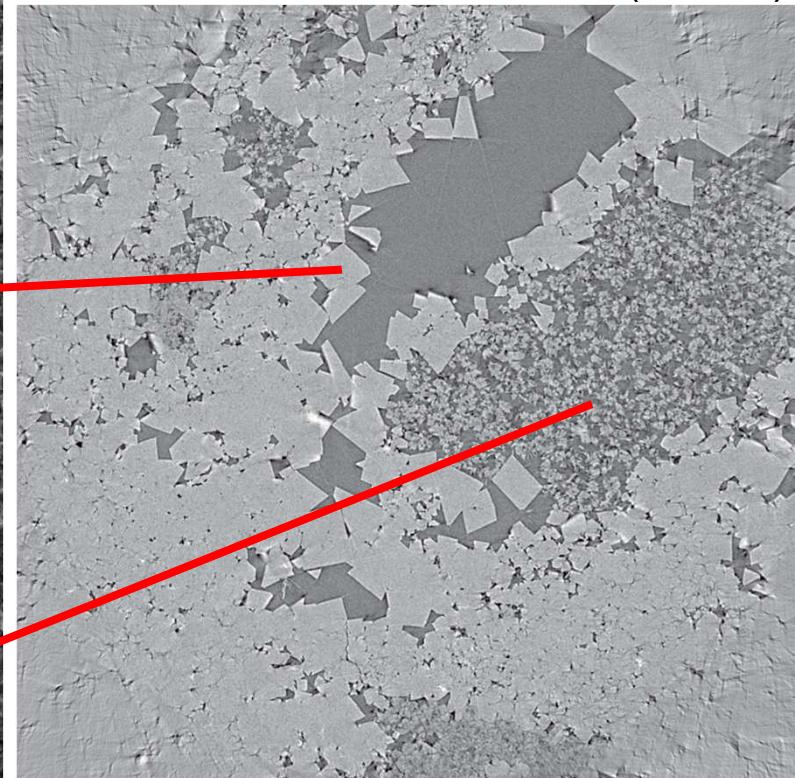
# Imaging the microporosity

SEM (825 m)



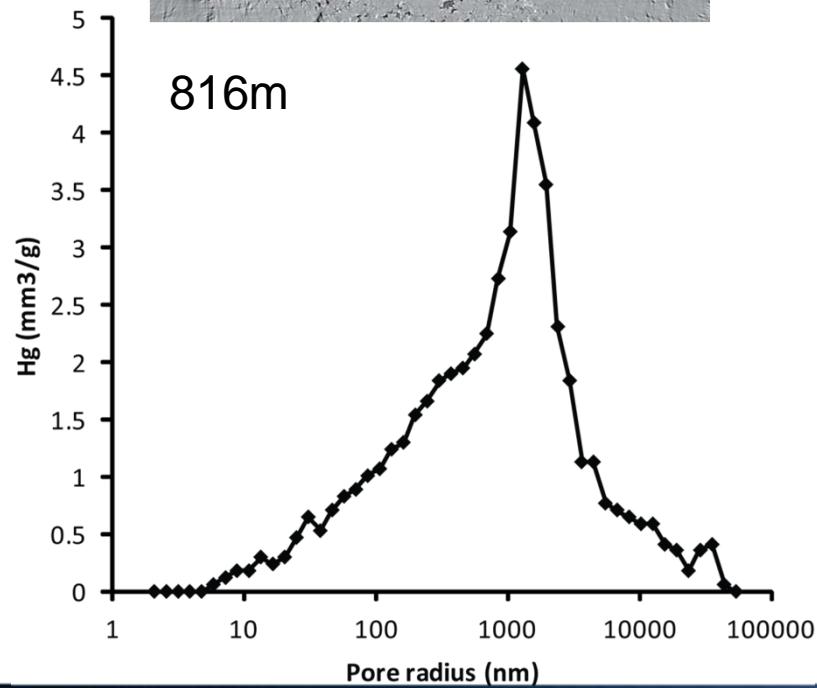
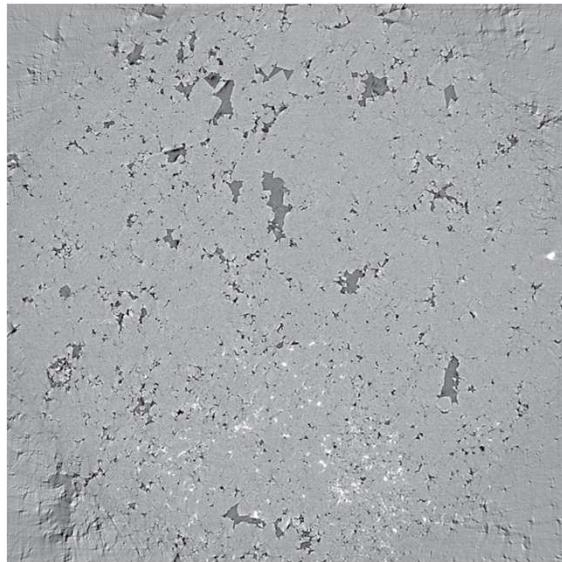
Scale bar = 20  $\mu\text{m}$

microCT (825 m)

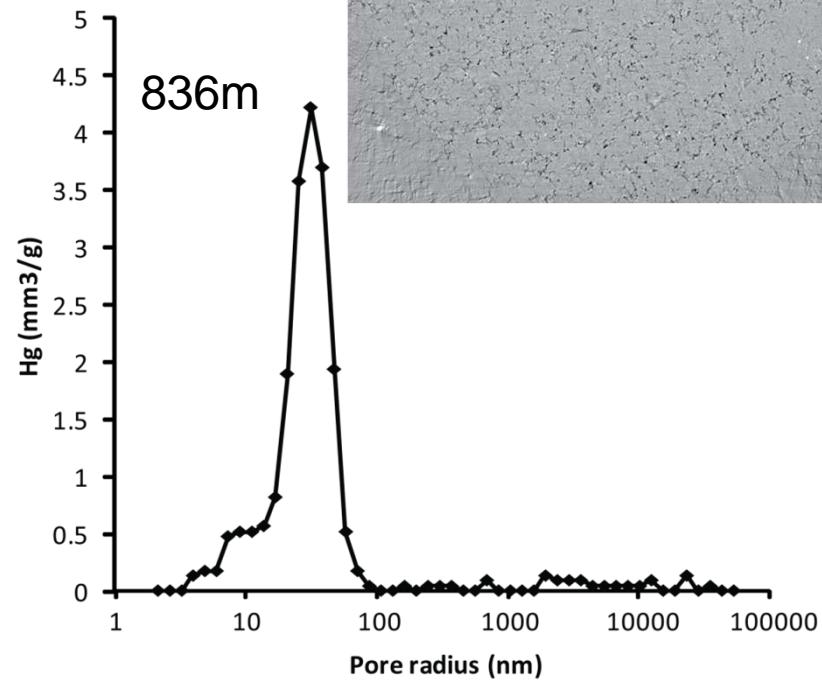
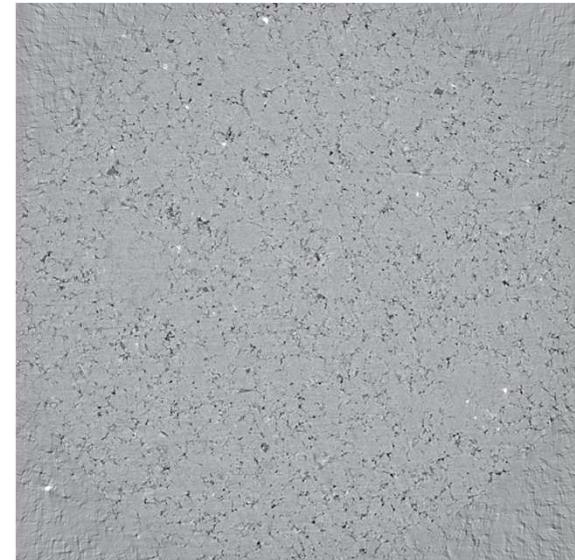


Viewfield = 750  $\mu\text{m}$   
Pixel Resolution = 0.38  $\mu\text{m}$

# The microporosity: mercury porosimetry



A significant portion of the microporosity is left out by the microCT measurements (<0.3 µm)



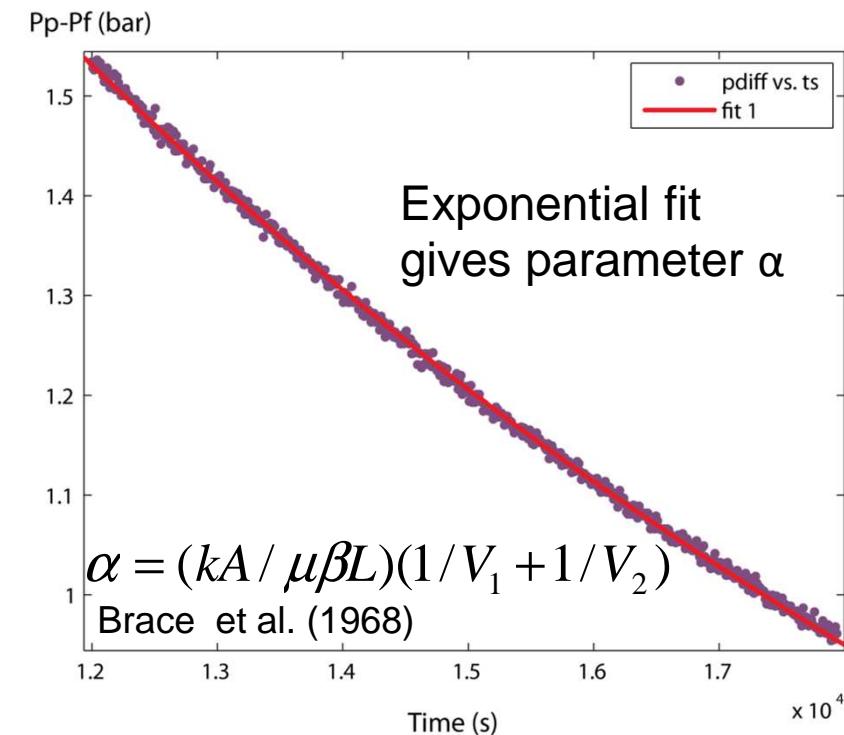
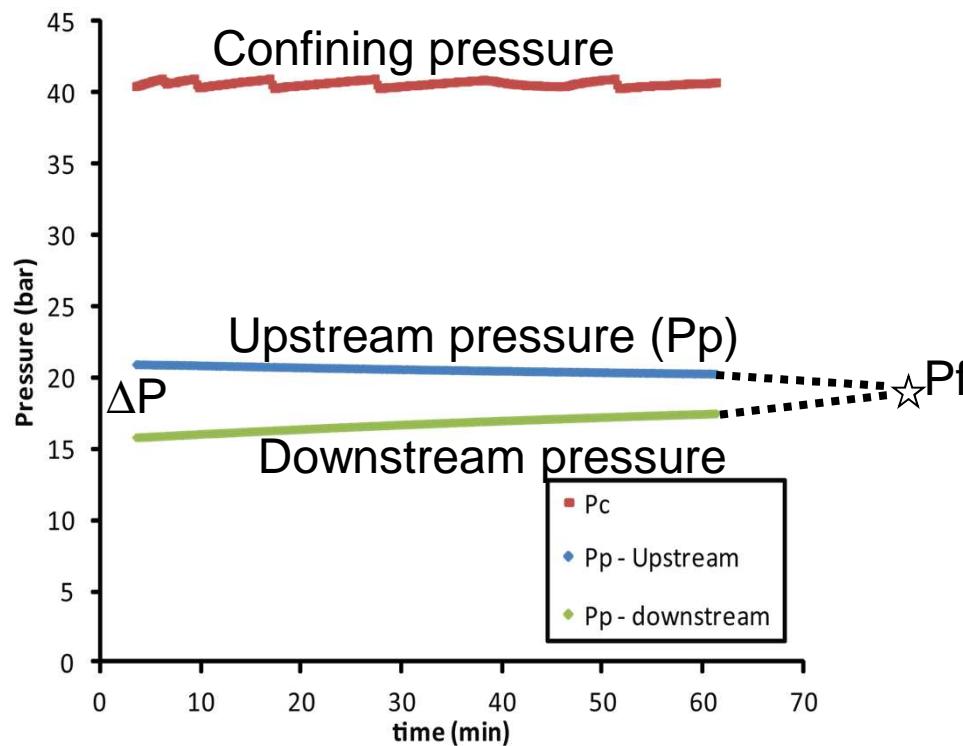
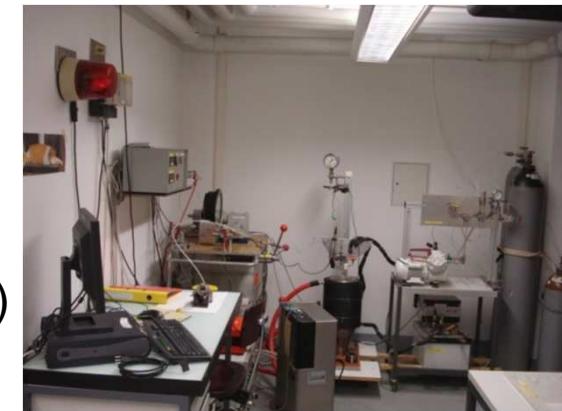
# Permeability measurements

Transient step technique

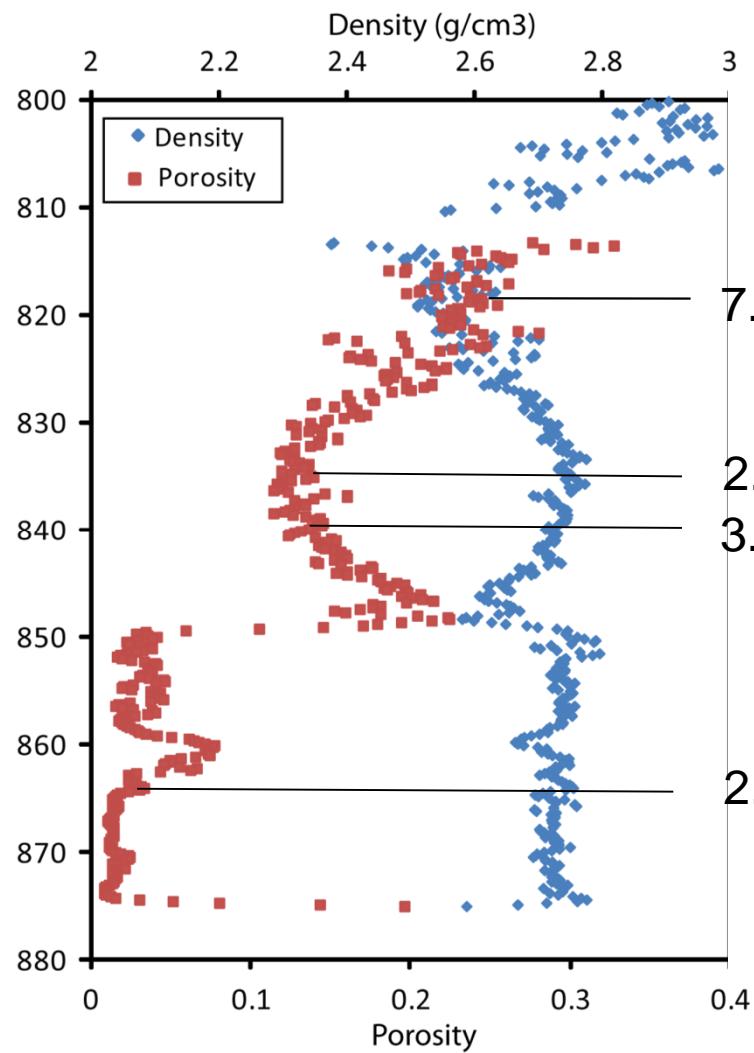
Confining pressure (300 bar) and temperature (up to 200 °C)

Pore fluid (gaseous, supercritical state, liquid)

1 inch diameter cores, 3-5 cm long



# Permeability measurements



Effective pressure: 120 – 140 bars  
Temperature: 45 °C  
Argon as pore fluid

7.16e-15 m<sup>2</sup> (817 m)

2.65e-18 m<sup>2</sup> (836 m)

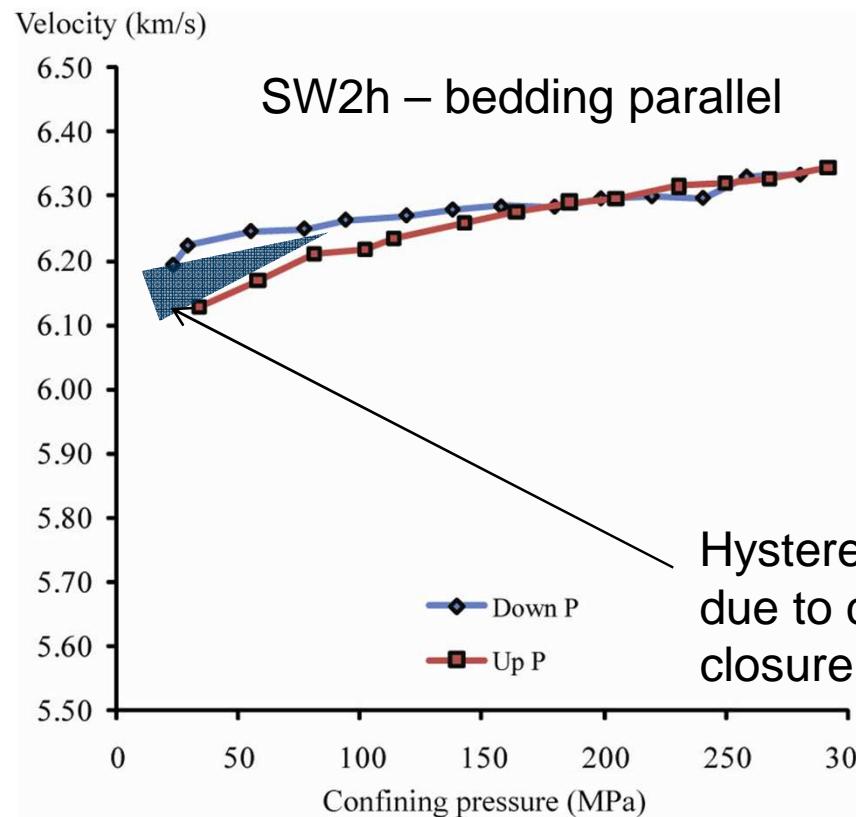
3.45e-18 m<sup>2</sup>

2.55e-19 m<sup>2</sup>

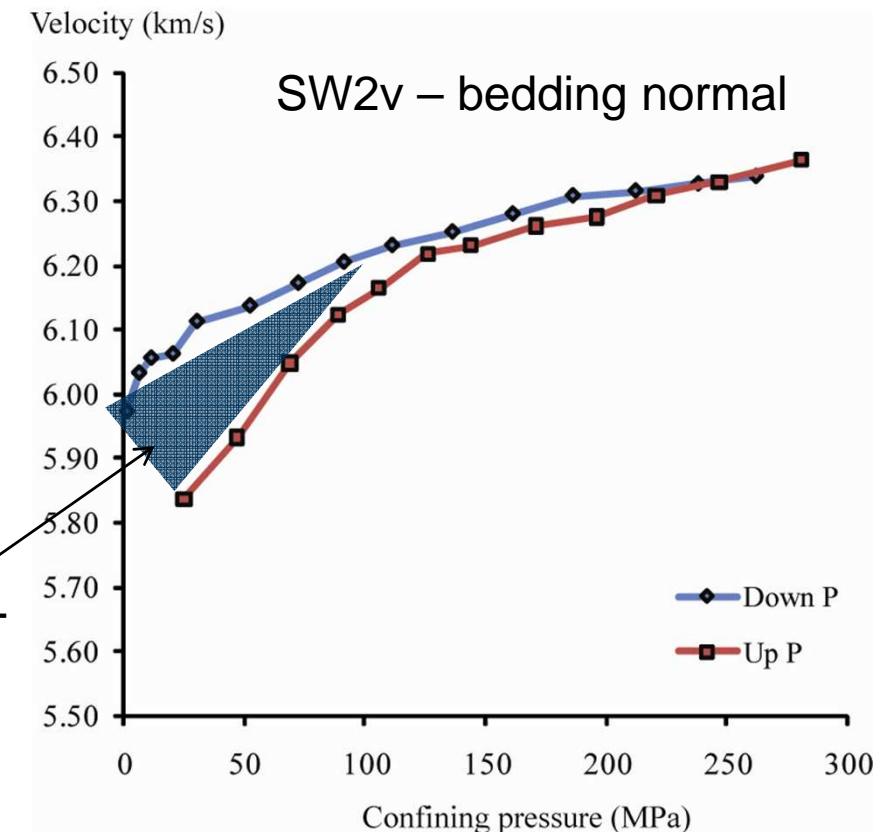
Samples chosen for permeability measurements show no large cracks (microporosity influence)

Measurements continue...

# Anisotropy of laboratory sonic velocity and permeability



Porosity = 3.9 %  
 $k = 1.14\text{e-}16 \text{ m}^2$  (130 bar Pe)

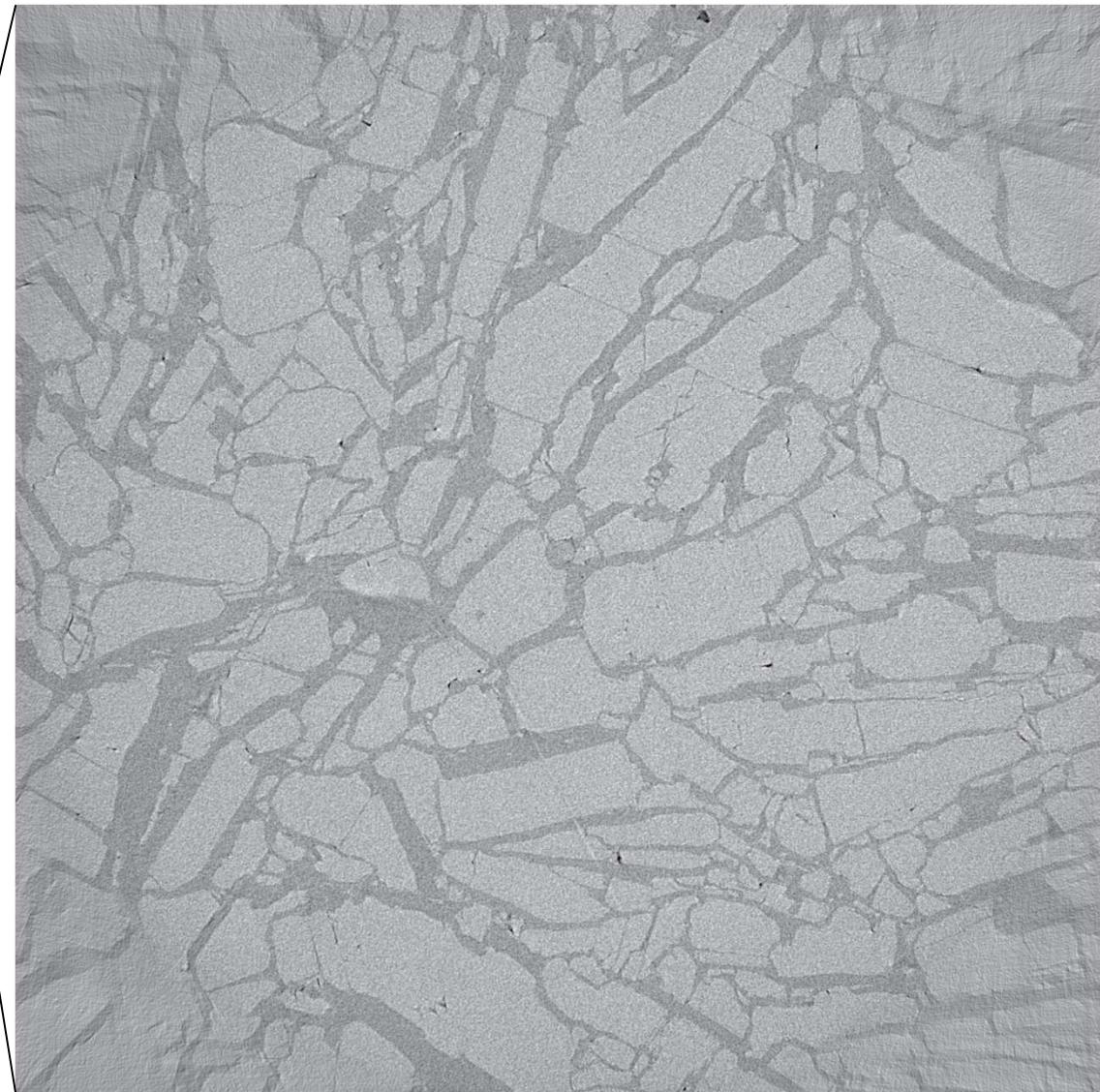


Porosity = 3.5 %  
 $k = 7.16\text{e-}18 \text{ m}^2$  (130 bar Pe)

# Conclusions

- Porosity and permeability in the upper Muschelkalk are heterogeneously distributed, over a large range:  
(<1% to 25% porosity;  $>10e-15 \text{ m}^2$  to  $\sim 10e-19 \text{ m}^2$ )
- Geochemical investigations indicate that important events in the development of porosity and permeability are related to diagenesis, as well as tectonic uplift and burial of the basin.
- The upper Muschelkalk is an aquifer, sometimes...
- What is the respective roles of the microporosity / macroscopic pores and cracks on the permeability?
- Upscaling:  
Lab <-> Borehole logging <-> Geophysical survey

# Thank you!



Viewfield = 750 µm