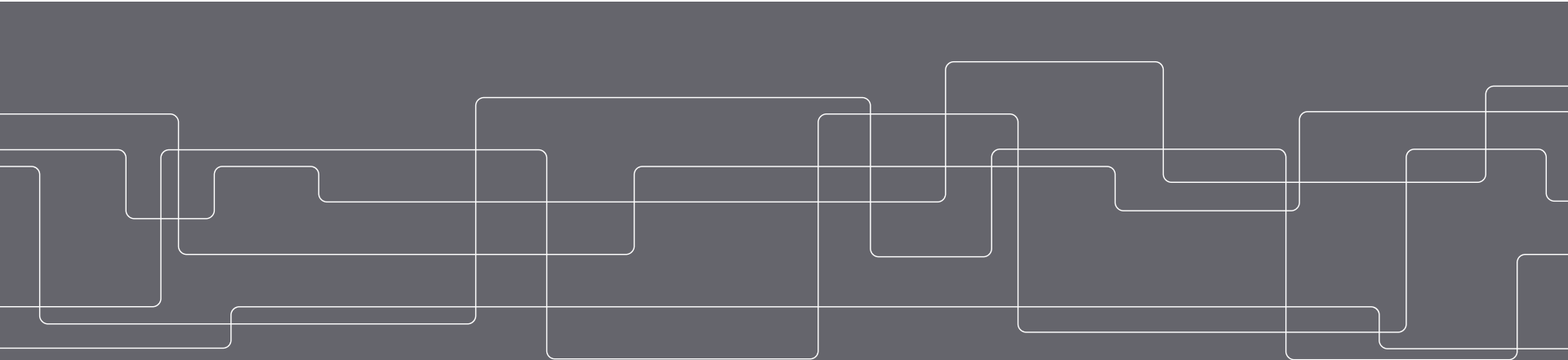




Ageing and Thermal Management

Henrik Ekström, Henrik Lundgren and Göran Lindbergh, Applied Electrochemistry, School of Chemical Science and Engineering, KTH

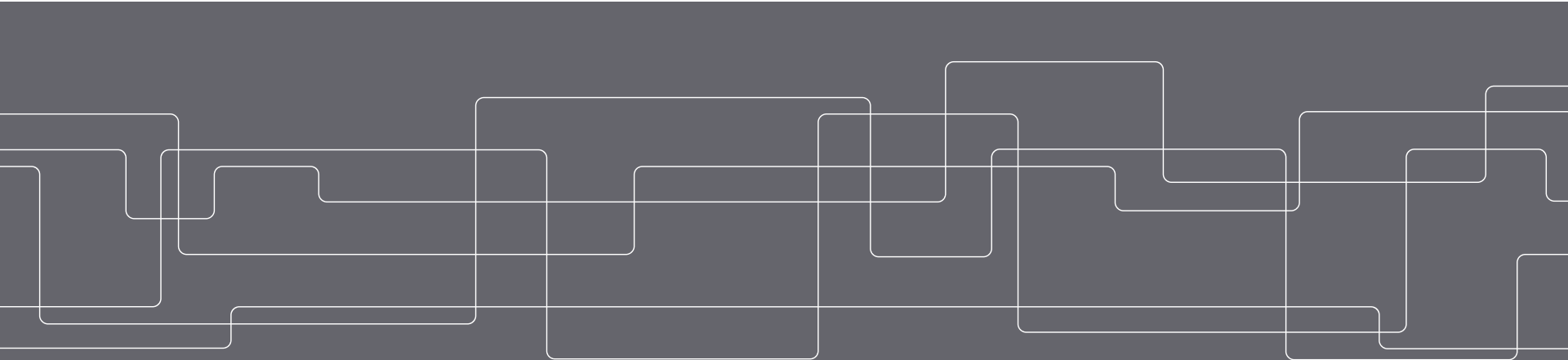
Nordbatt2, Trondheim, 2nd dec 2015



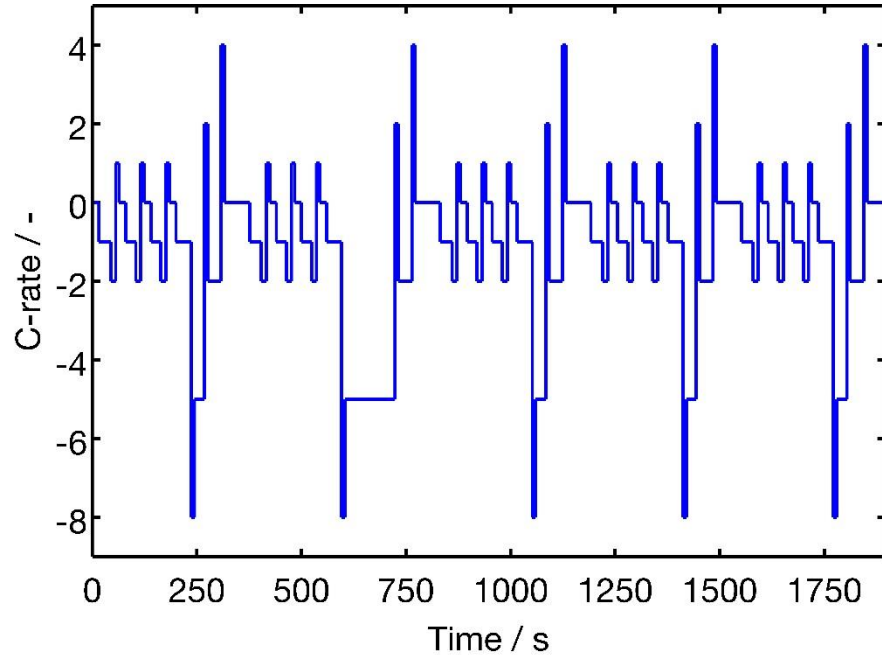


Thermal Battery Characterization

See also: Henrik Lundgren et al, Journal of The Electrochemical Society, 163 (2) A1-A9 (2016)



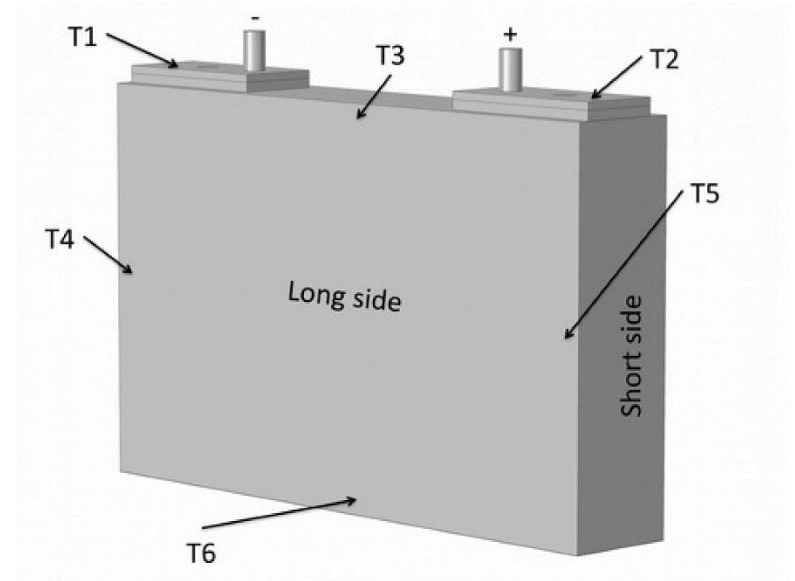
What is the Internal Battery Temperature?



Hybrid Cycle used
at Scania CV

Experimental Setup

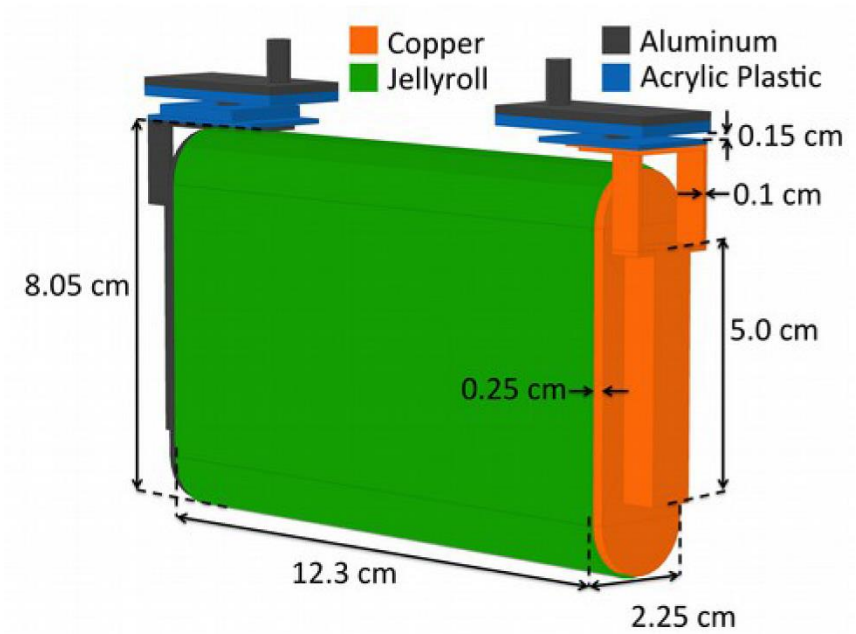
- Large format 25 Ah NMC/graphite cell (DIN SPEC 91252:2011-01)
- Temperature Probes
- Heat sink at 25 °C, placed at bottom
- Hybrid load cycle up to 8C



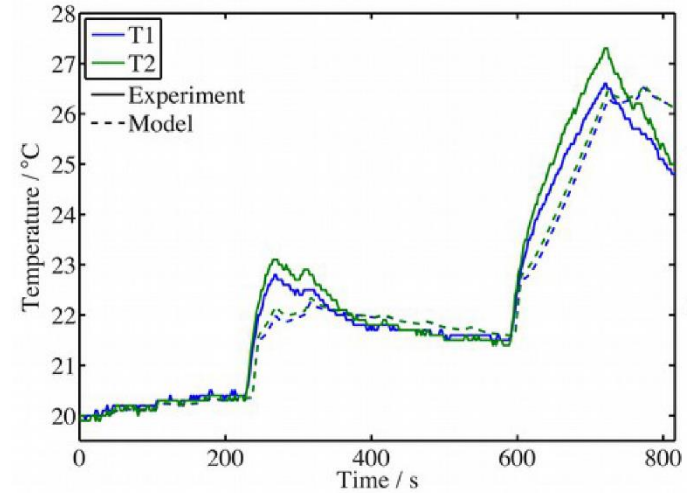
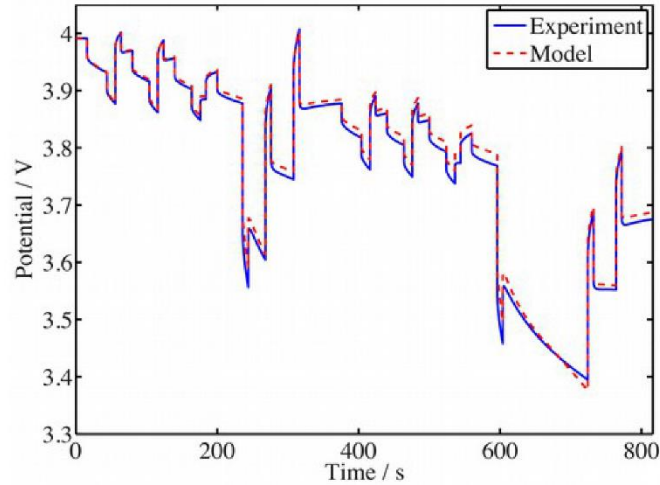
Temperature probe positions

Model Setup

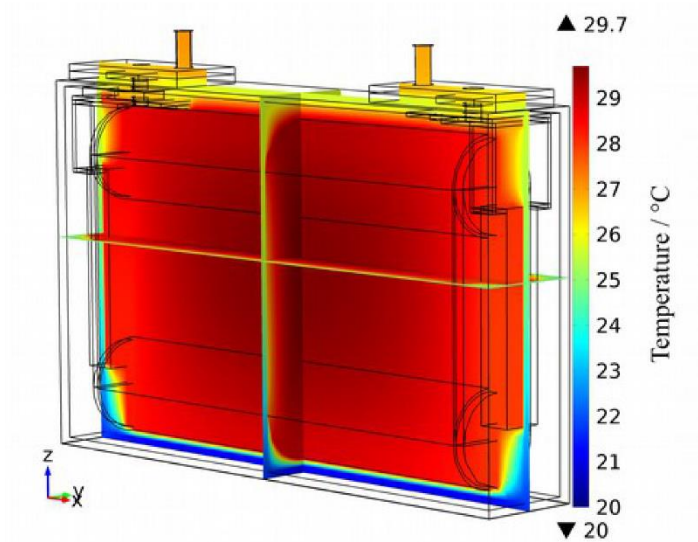
- Heat transfer model
 - Full 3D geometry
 - Joule heating in current collectors
 - Electrochemistry heat sources from electrochemical model in jelly roll
- Electrochemical battery model
 - Doyle/Newman model with reduced dimension to 1D(+1D) using average jelly roll temperature



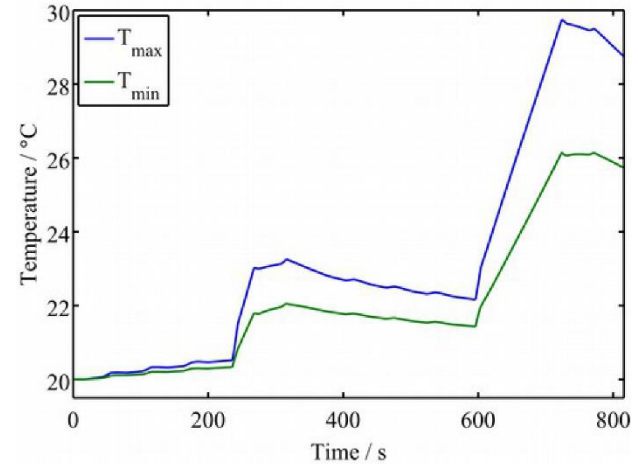
Model Validation



Internal Temperatures



Internal Temperatures at 724 s



Max/Min Jelly Roll Temperatures



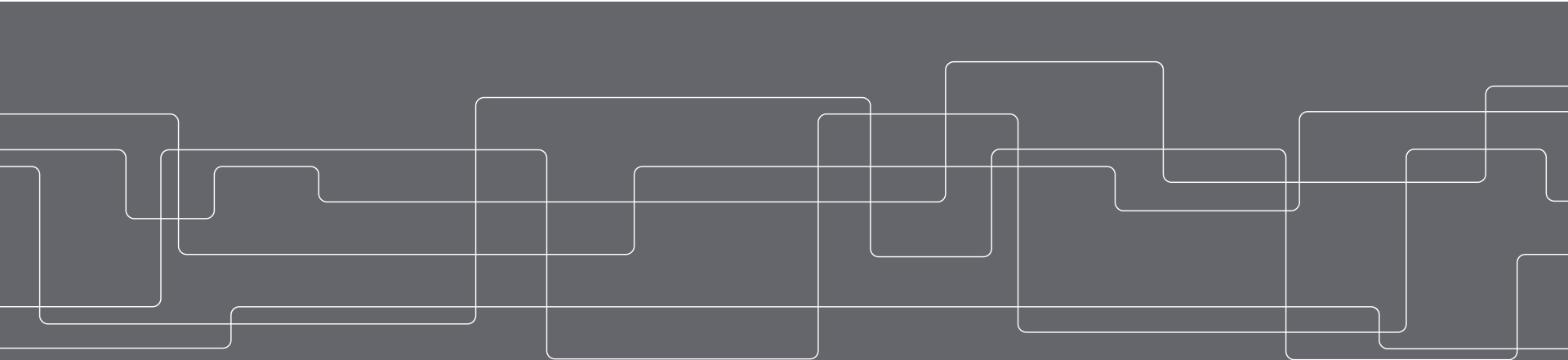
Conclusions from Thermal Characterization

- Orientation matters - long side cooling will cause higher internal gradients
- Main path for heat flux is between jelly roll and can – not through current collectors
- Largest internal temperature differences $< 5\text{ }^{\circ}\text{C}$
 - Temperature-induced local differences in current density and state-of-charge are expected to be small
 - Temperature-induced non-uniform ageing effects would be expected, but small



Battery Ageing Modelling

See also: Henrik Ekström and Göran Lindbergh, *Journal of The Electrochemical Society*, 162 (6) A1003-A1007 (2015)



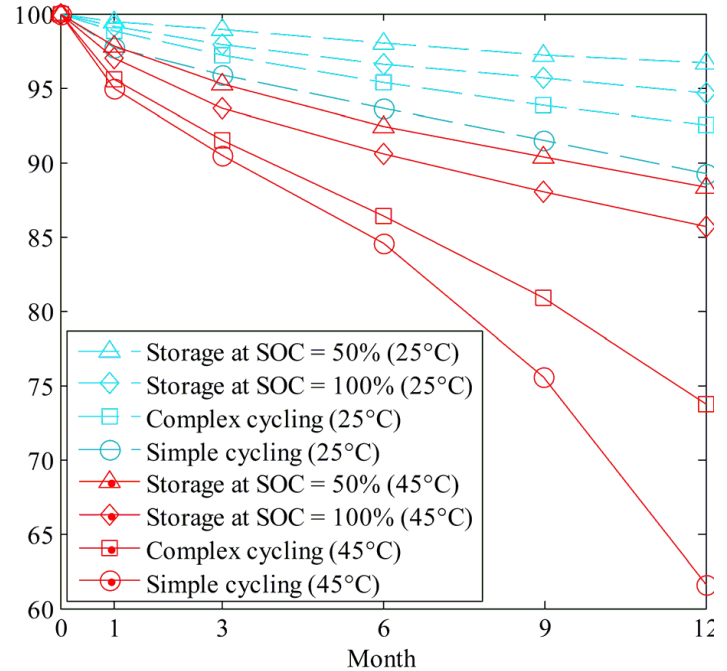


Experimental Data: LFP/Graphite 18650 cell

Ageing rate depends on:

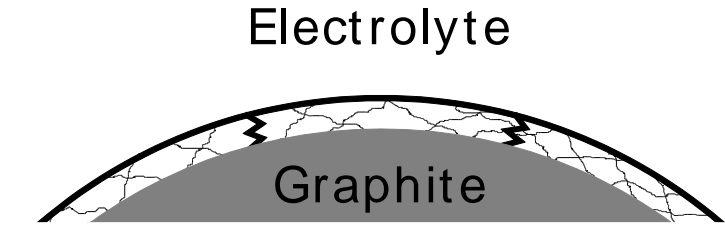
- State-of-charge
- Current
- Temperature
- History

Data: Safari and
Delacourt, JES 158
A1123 (2011)



SEI Layer Growth Model

- SEI growth due to reduction of an electrolyte species
- Transport occurs through covering and cracked regions of the layer
- SEI forming electrode kinetics depends both on potentials and layer transport limitations
- Cracked area is proportional to graphite expansion rate



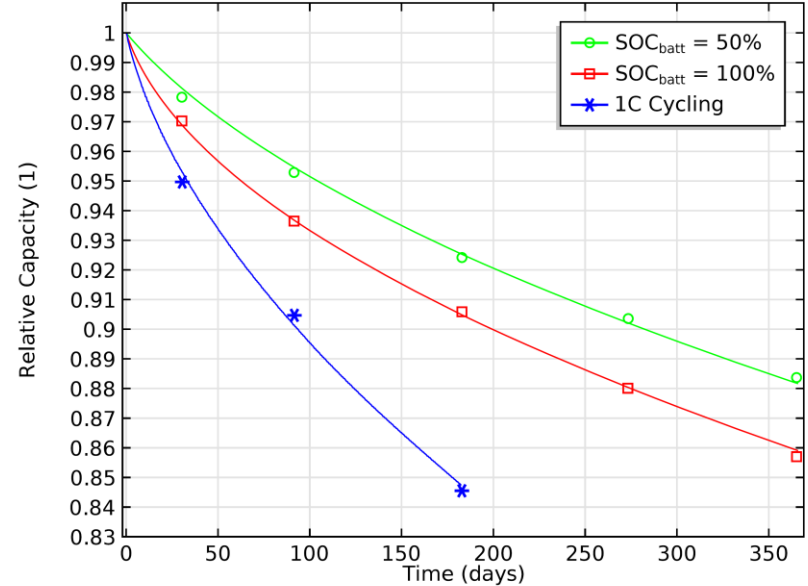
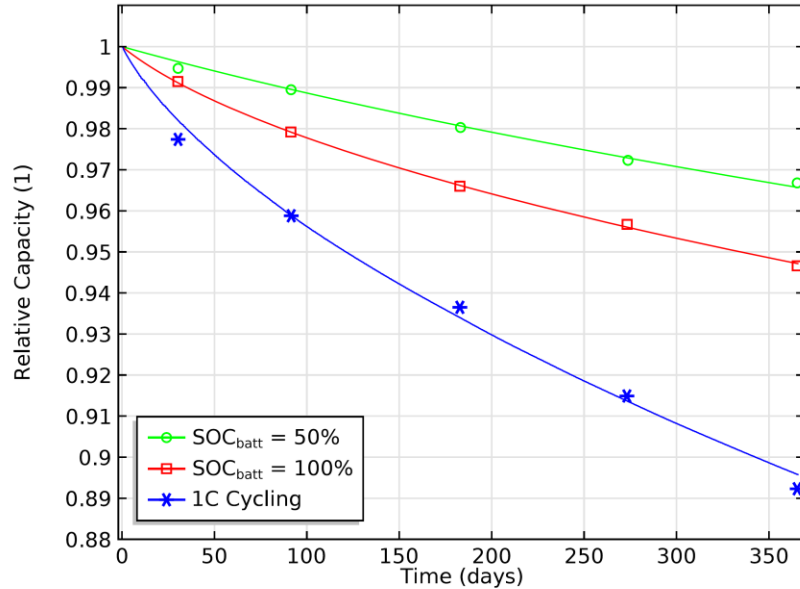


SEI Equation and Model Parameters

$$I_{\text{SEI}} = -(1 + HK_{\text{crd}}) \frac{JI_{1\text{C}}}{\exp\left(\frac{\alpha\eta_{\text{SEI}}F}{RT}\right) + \frac{Q_{\text{SEI}}fJ}{I_{1\text{C}}}}$$

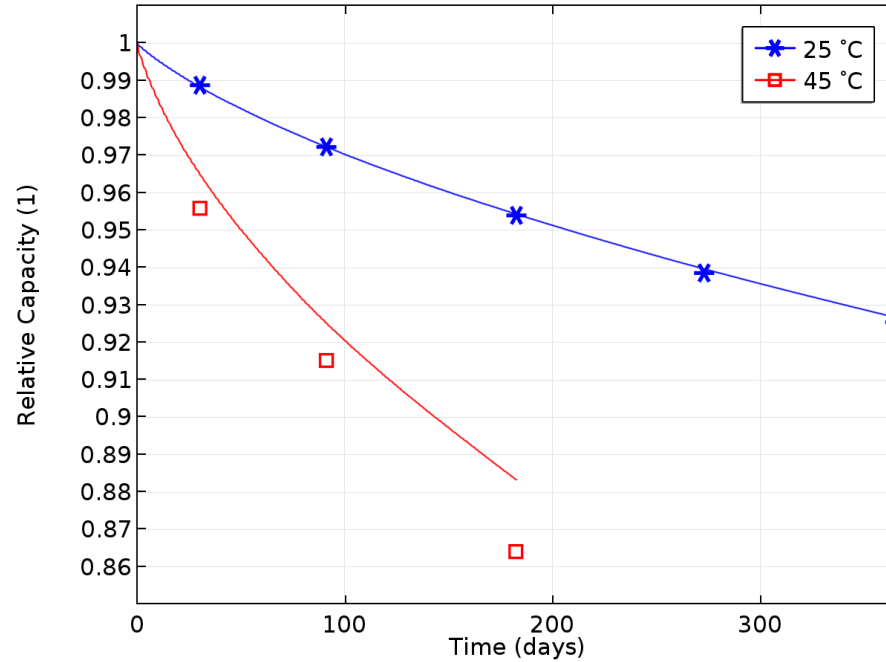
- α – electrokinetic transfer coefficient
- J – lumped kinetic parameter
- f – lumped SEI layer diffusion limitation parameter
- H – lumped crack area
- One set of four parameters needed for each temperature

Model Fitting Results at 25°C and 45°C





Model Prediction Tests





Future and Ongoing Work

- Temperature Characterization
 - Temperature probes inside cells
 - Temperature controlled current cables
- Ageing
 - Single electrode experiments
 - Graphite electrode ageing models at higher currents
 - ...what would be the effect of the high current pulse?
 - Positive electrode materials
 - ... what is the effect of the NMC positive electrode?

Finally, some homework for you...

...we need better temperature dependent data on

- capacity and power fade
- heat conductivities
- heat capacities
- electrode kinetic parameters
- equilibrium (open-circuit) potentials
- lithium diffusion in the active materials