



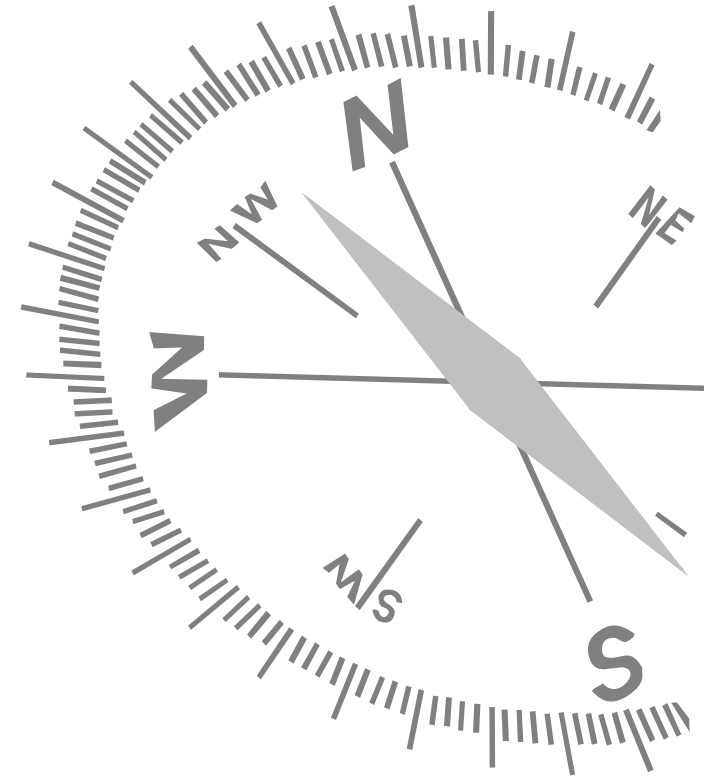
MARITIME ACADEMY

# (Future) Fuels & Fuel Converters

# Navigator

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- ➔ Fuel types
- Natural gas
- Fuel cells
- Quiz



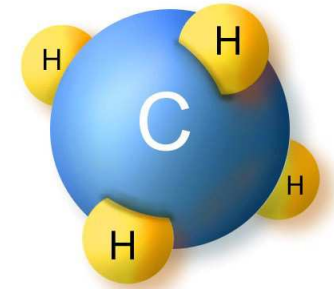
## What is a “fuel”?

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- Fuel:**    **Substance**                      which in
- **chemical reaction**                with an
  - **oxidizer**                                (typically oxygen) releases
  - **heat**



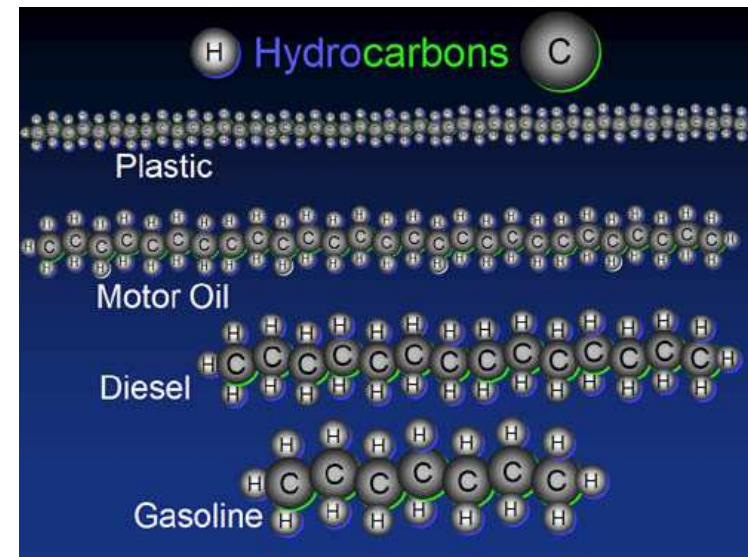
## Carbon-based fuels



Most fuels contain carbon-hydrogen compounds

Longer chains:

- **heavier** fuel
- more **viscous** fuel
- **lower calorific** value
- **higher temperature** to evaporate



## Fuels – come in different forms

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Fuels may be

- solid
- liquid
- gaseous



# Fossil vs Renewable

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Fuels may be

fossil

OR

renewable

coal

wood

petroleum-crude derived

refuse

natural gas

agricultural residues

bio fuels



## Case Study – Fuel & Energy

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Rank fuels in energy obtained when 1 g of the fuel is burnt

- alcohol  $(\text{C}_2\text{H}_5\text{OH})$
- butane gas  $(\text{C}_4\text{H}_{10})$
- solid carbon  $(\text{C})$
- glucose sugar  $(\text{C}_6\text{H}_{12}\text{O}_6)$
- hydrogen gas  $(\text{H}_2)$



# Place 5



# Place 4

# Bronze

# Calorific value

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**Silver**

## Calorific value

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**Gold**

## Calorific values

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<b>Sugar</b>	<b>18 kJ/g</b>	
<b>Alcohol</b>	<b>30 kJ/g</b>	
<b>Coal</b>	<b>30 kJ/g</b>	
<b>HFO</b>	<b>41 kJ/g</b>	Heavy Fuel Oil
<b>MDO</b>	<b>44 kJ/g</b>	Marine Diesel Oil
<b>MGO</b>	<b>45 kJ/g</b>	Marine Gas Oil
<b>Butane</b>	<b>50 kJ/g</b>	main component of LPG
<b>Carbon</b>	<b>53 kJ/g</b>	
<b>Hydrogen</b>	<b>143 kJ/g</b>	good for storing energy (e.g. from wind)

LPG = Liquid Petroleum Gas

## Marine fuels in use

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- MGO (Marine Gas Oil)      roughly equivalent to No. 2 fuel oil  
made from distillate only
- MDO (Marine Diesel Oil)      blend of gasoil and heavy fuel oil
- IFO (Intermediate Fuel Oil)      blend of gasoil and heavy fuel oil  
(less gasoil than MDO)
- MFO (Medium Fuel Oil)      blend of gasoil and heavy fuel oil  
(even less gasoil than IFO)
- HFO (Heavy Fuel Oil)      (nearly) pure residual oil  
roughly equivalent to No. 6 fuel oil

## Common fuels

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Viscosity of fuels (ordered by increasing price):

- IFO 380      intermediate fuel oil with  $\nu \leq 380$  cSt
- IFO 180      intermediate fuel oil with  $\nu \leq 180$  cSt
- LS 380      low-sulphur (<1.5%) IFO with  $\nu \leq 380$  cSt
- LS 180      low-sulphur (<1.5%) IFO with  $\nu \leq 180$  cSt
- MDO          Marine diesel oil                      < 30 cSt at 50°C
- MGO          Marine gas oil                              < 6 cSt at 40°C

# Viscosity

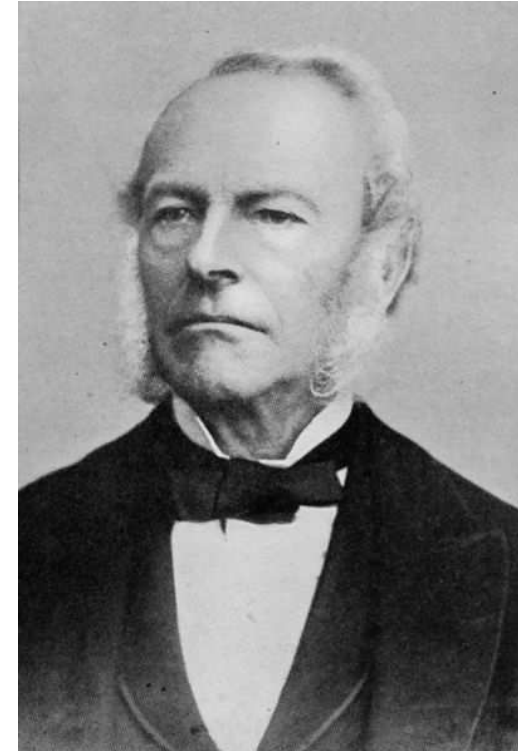
Kinematic viscosity:

$$v = \frac{\mu}{\rho}$$

$\mu$  = dynamic viscosity [Pa·s]  
 $\rho$  = density [kg/m<sup>3</sup>]  
 $v$  = kinematic viscosity [m<sup>2</sup>/s]

1 stokes [St] = 1 cm<sup>2</sup>/s = 0.0001 m<sup>2</sup>/s

1 centistokes [cSt] = 1 mm<sup>2</sup>/s = 10<sup>-6</sup> m<sup>2</sup>/s



Sir George Gabriel Stokes,  
1<sup>st</sup> Baronet (1819–1903)



milk 4.3 cSt

just like...

SAE 20 Crankcase Oil

SAE 75 Gear Oil



tomato juice    220 cSt

just like...

SAE 50 Crankcase Oil

SAE 90 Gear Oil



honey 2200 cSt



## Questions so far?

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# Navigator

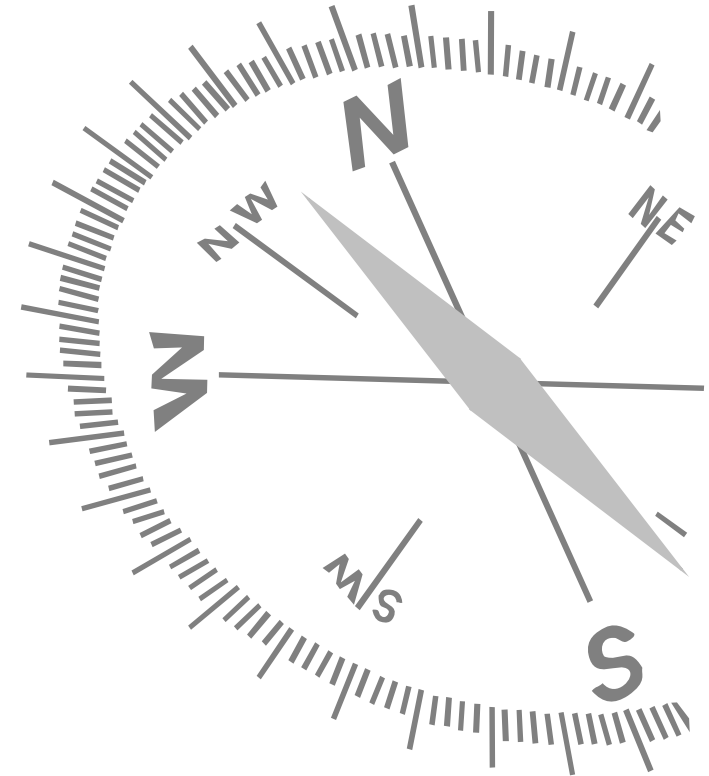
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Fuel types

➔ Natural gas

Fuel cells

Quiz



## What is “natural gas” ?

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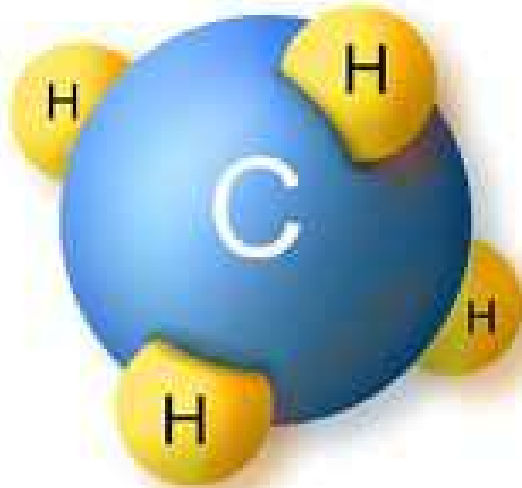
- Composition of different gases
- Actual mix depends on gas field (and processing of gas)



## What is the main component in natural gas?

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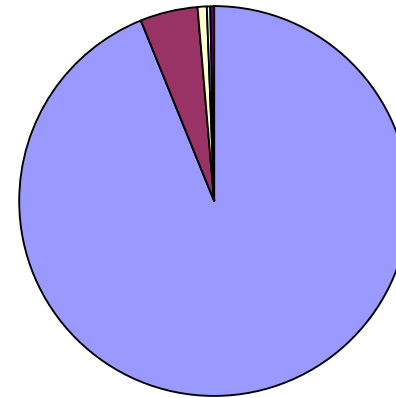
What is the “usual” name of the shown substance ?



## (Liquefied) natural gas – Methane dominates

Typical composition in volume %

Methane	94.0 %
Ethane	4.7 %
Propane	0.8 %
Butane	0.2 %
Nitrogen	0.3 %



Density: 0.716 kg/m<sup>3</sup> at 273 K (0°C) and ambient pressure



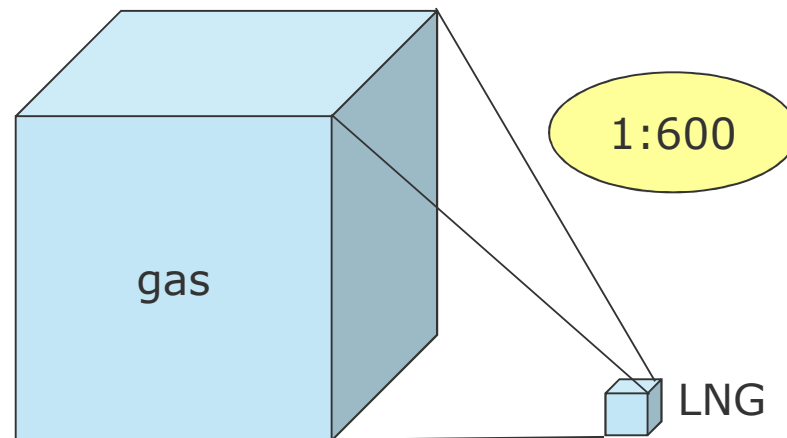
## Liquid gas takes up much less space

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Volume ratio liquid : gas (LNG) = 1:600

(1 bar,  $-163^{\circ}\text{C}$ :  $\rho = 425.0 \text{ kg/m}^3$ )

(1 bar,  $0^{\circ}\text{C}$ :  $\rho = 0.7 \text{ kg/m}^3$ )



## LNG vs HFO

For the storage of 1 t of LNG, you need...



- a. roughly the same...
- b. roughly twice the ...
- c. roughly five times the ...
- d. roughly ten times the ...

...volume as for 1 t of HFO ?

LNG = Liquefied Natural Gas  
HFO = Heavy Fuel Oil

## LNG vs HFO

For 1 m<sup>3</sup> of LNG, you get



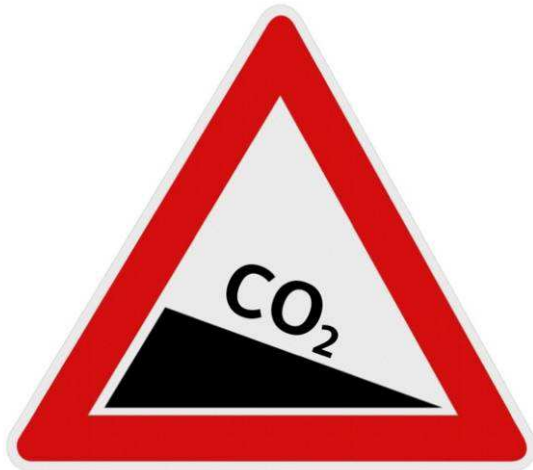
- a. roughly 10% less than the ...
- b. roughly the same ...
- c. roughly 10% more than the ...
- d. roughly 20% more than the ...

... heat for 1 m<sup>3</sup> of HFO ?

## LNG vs HFO

Compared to HFO, LNG decreases CO<sub>2</sub> emissions by roughly...

- a. 10% ...
- b. 25% ...
- c. 40% ...
- d. 90% ...



... for the same work [g CO<sub>2</sub>/MJ]

## Cause the fuels, they are a-changing

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Over time, the “standard” (or predominant) fuel has changed

Bring the fuels (below in alphabetical order) in correct order of time from medieval times to “future”

- coal
- hydrogen
- natural gas
- oil
- wood



## Sort again – by hydrogen content

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All fossil fuels contain hydrogen and carbon

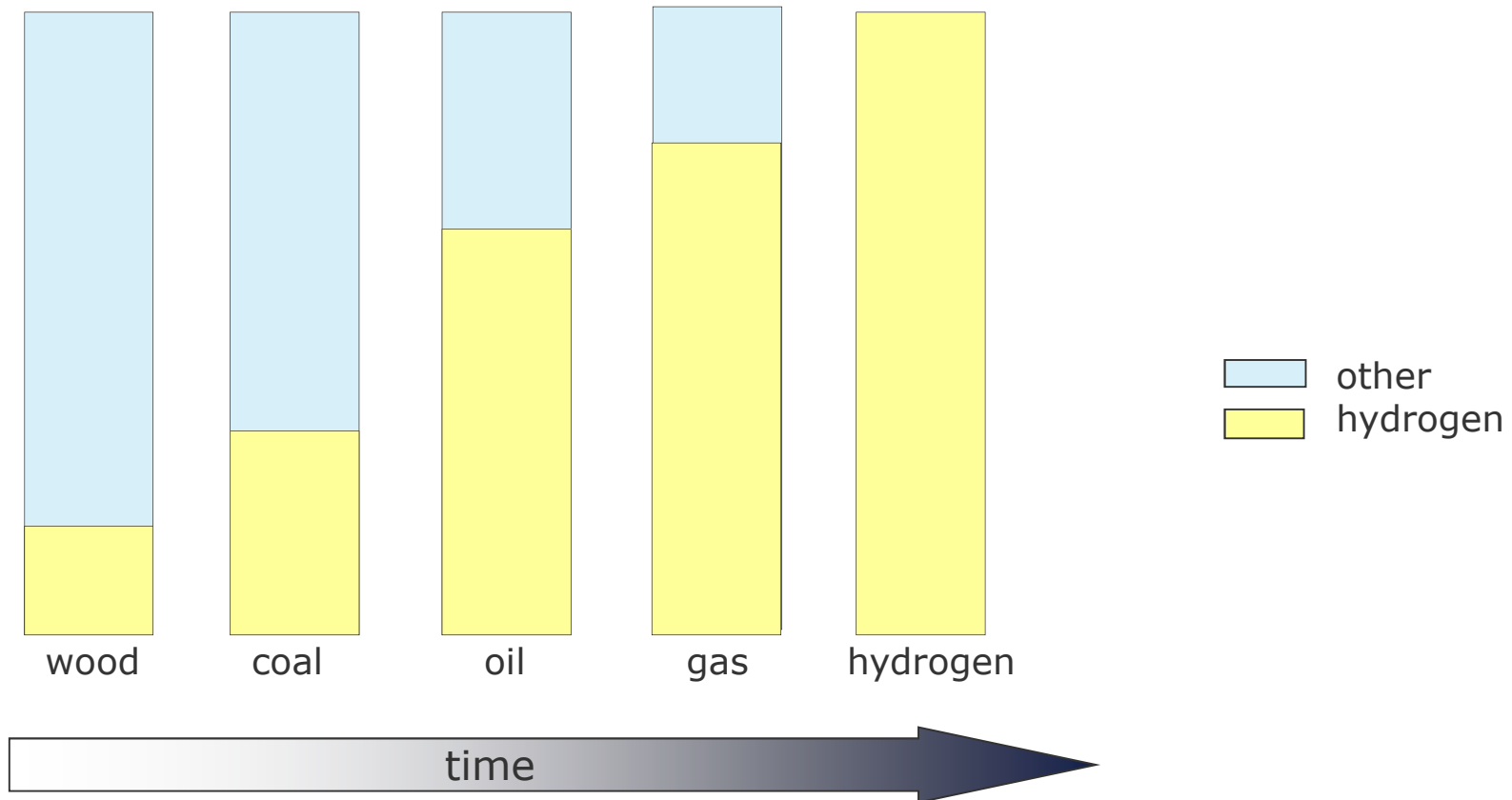
Bring the fuels (below in alphabetical order) in correct order of increasing hydrogen content

- coal
- hydrogen
- natural gas
- oil
- wood



## Fuels have changed over time

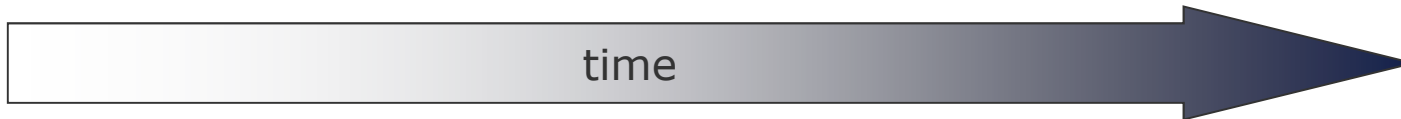
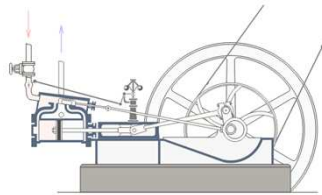
Carbon content in fuel



## Different strokes for different fuels

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Engines (Energy converters) change with time (and fuel)

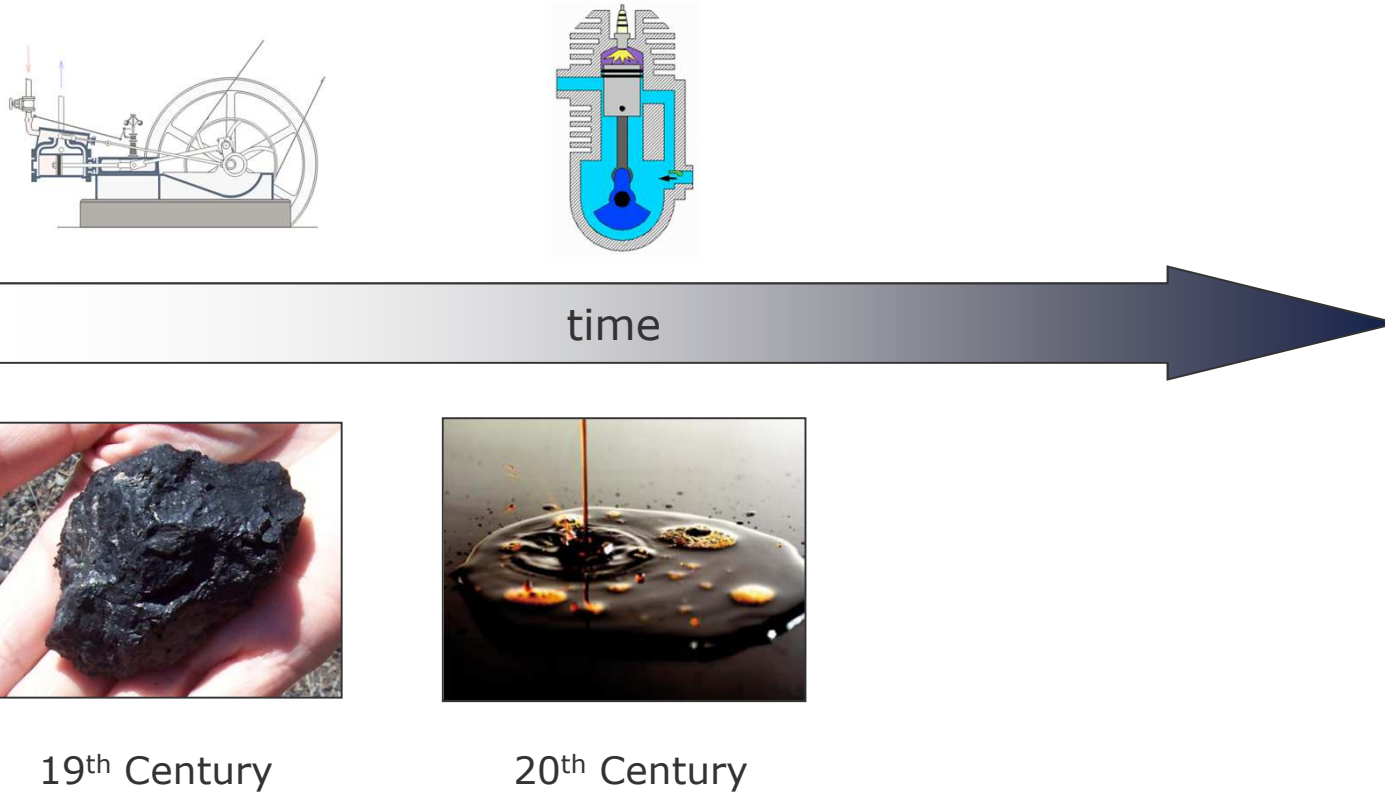


19<sup>th</sup> Century



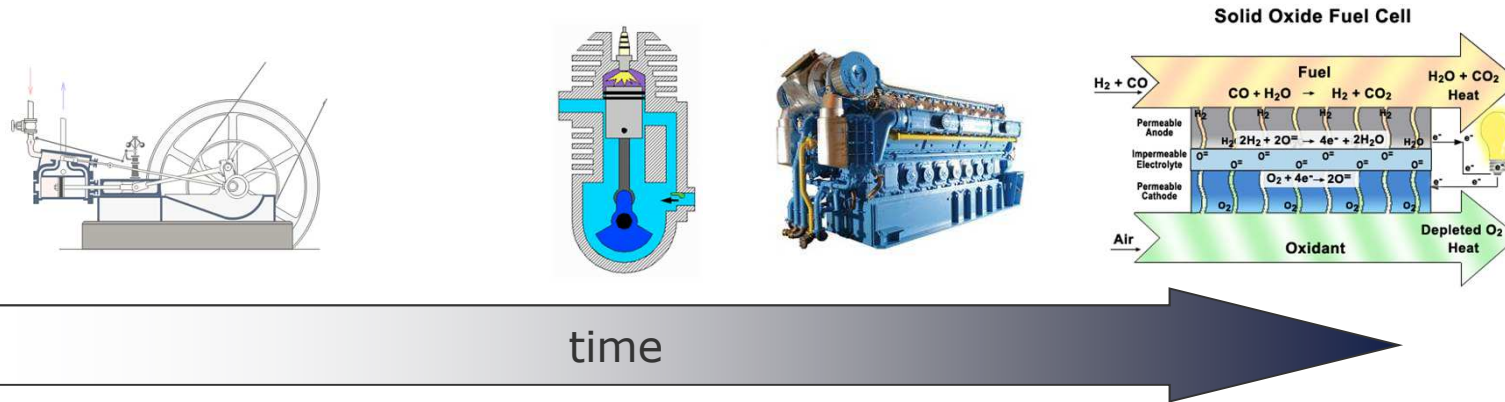
## Different strokes for different fuels

Engines (Energy converters) change with time (and fuel)



# Different strokes for different fuels

Engines (Energy converters) change with time (and fuel)



19<sup>th</sup> Century



20<sup>th</sup> Century



21<sup>th</sup> Century



## Questions so far?

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# Navigator

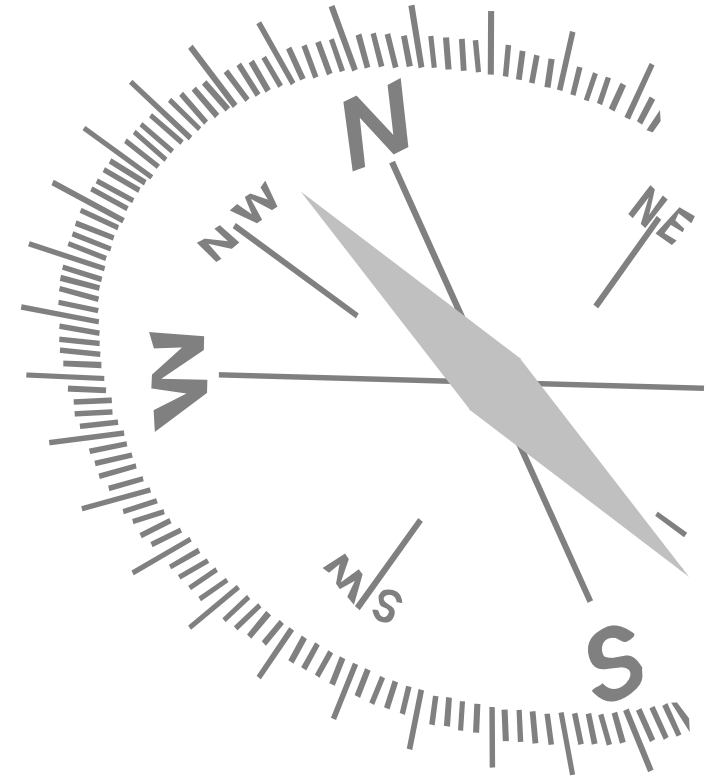
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Fuel types

Natural gas

 Fuel cells

Quiz



Let's start with something similar, but much more familiar



An electro-chemical energy conversion device ...  
... or simply: "**battery**"

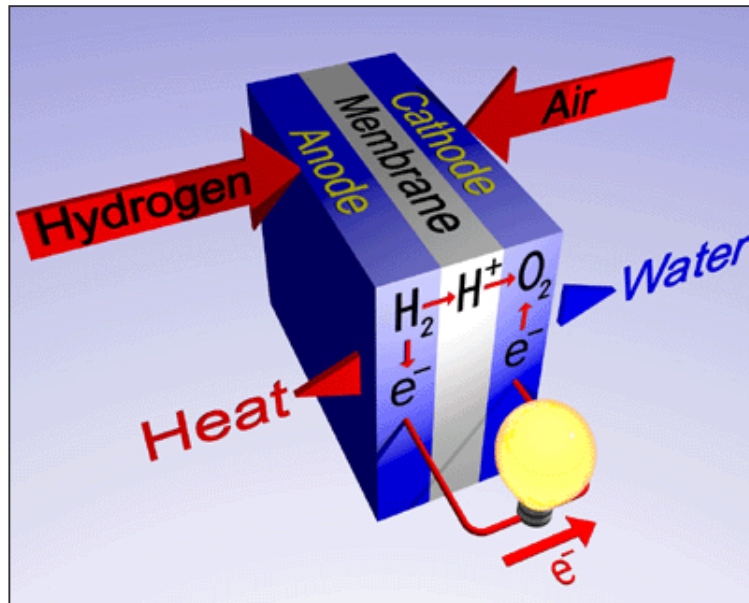
- all chemicals stored inside
- converts the chemicals to electricity
- will eventually "go dead"

## What is a fuel cell ?

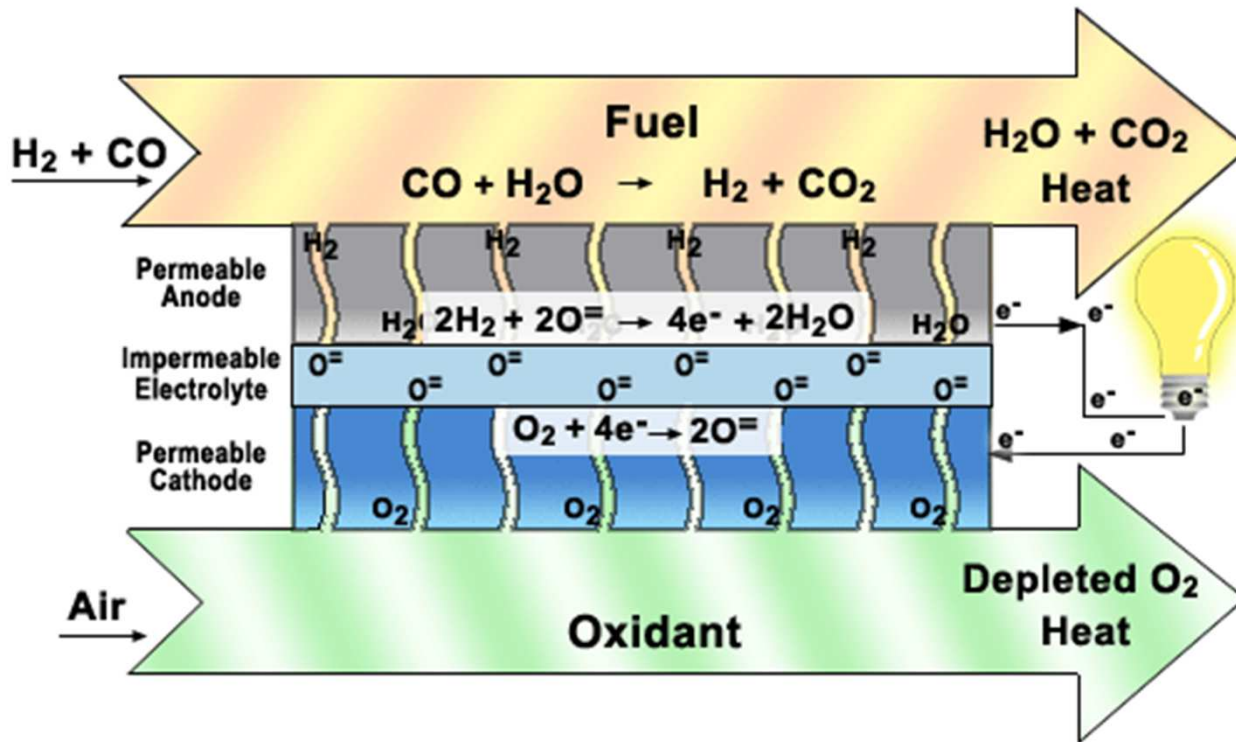
2/2

Also an electro-chemical energy conversion device

- Chemicals constantly flow into the cell (it never “goes dead”).
- Most fuel cells use hydrogen & oxygen, producing water & electricity.



# Fuel cell in action



Source: NASA

## Many variations on the theme

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Fuel cell types differ by:

- electrolyte (main classifier)
  - PEM (Proton Exchange Polymer)
  - PAFC (Phosphoric Acid Fuel Cell)
  - MCFC (Molten Carbonate Fuel Cell)
  - SOFC (Solid Oxide Fuel Cell)
- fuel (hydrogen, methanol, ethanol, natural gas, carbon-monoxide, ...)
- oxidant (usually oxygen)
- temperature
  - low temperature (< 100°C)
  - high temperature (600-1100°C)



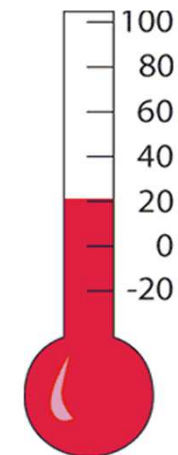
## Cool and fast...

### Low-temperature fuel cells

- ☺ rapid start-up
- ☹ requires hydrogen or methanol as fuel
- ☹ catalysts easily poisoned
- ☹ low efficiency

#### Applications:

- portable devices
- frequent on/off cycles
- compact devices



## Hot and slow...

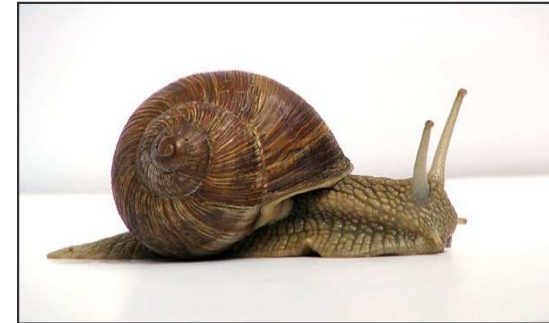
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### High-temperature fuel cells

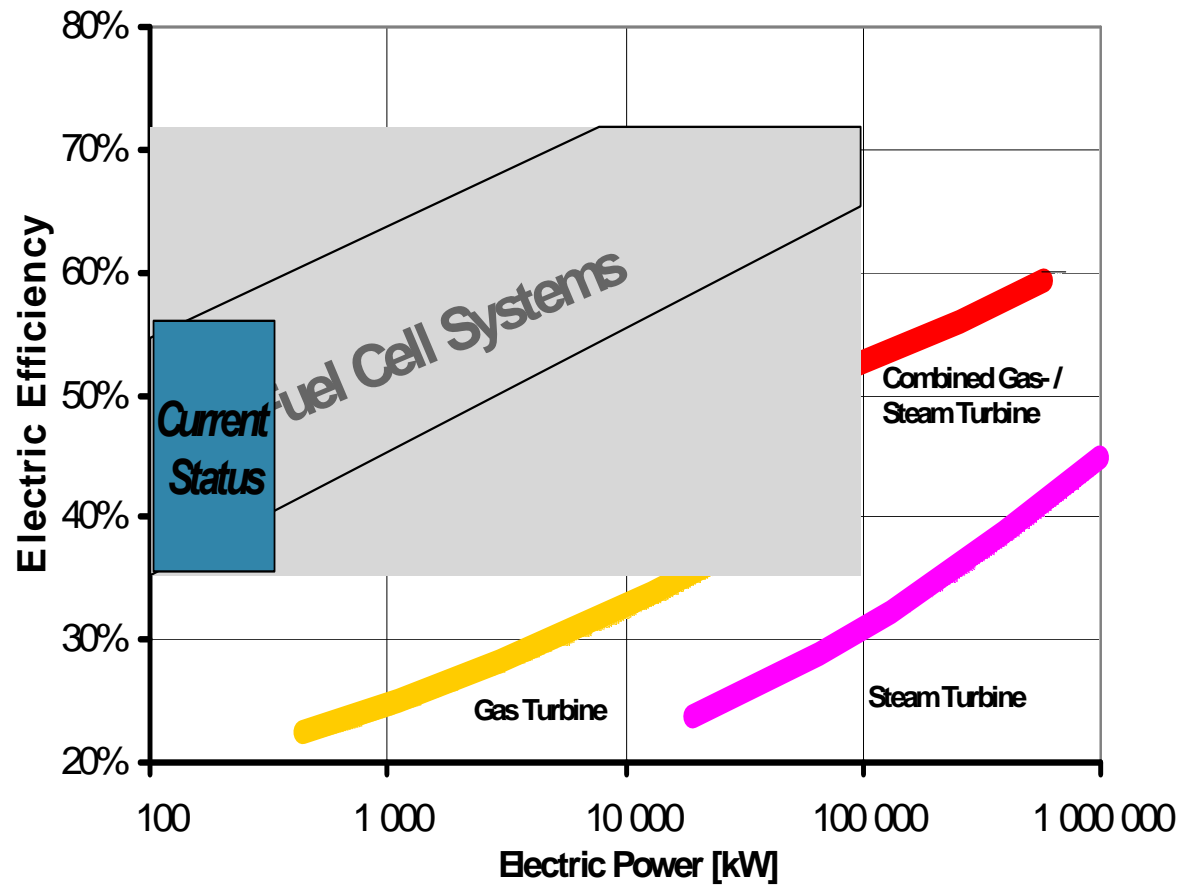
- ☺ fuel flexible
- ☺ high efficiency
- ☹ long start-up

Applications:

- stationary power
- ships



## Higher efficiency than diesel engines



## History of fuel cells

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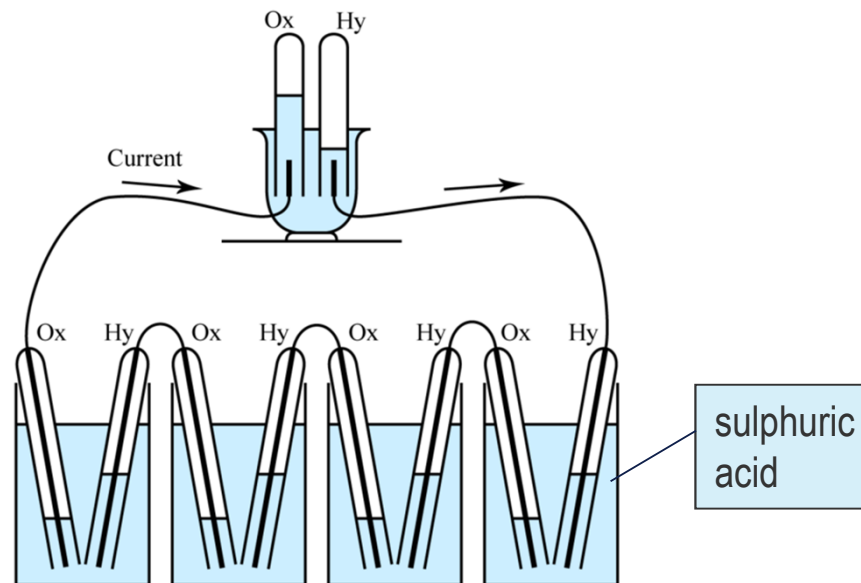
- 1838** Christian Friedrich Schönbein (Germany)  
chemist  
discovers principle of fuel cells  
(using two platinum wires and sulphuric acid)  
discovered also guncotton and ozone



## History of fuel cells

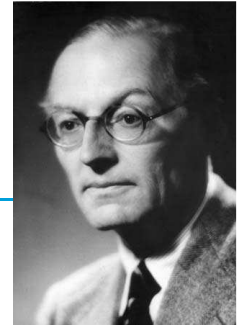


**1839** Sir William Grove (UK)  
lawyer & physical scientist  
first working prototype  
voltage  $\sim 1$  V



## History of fuel cells

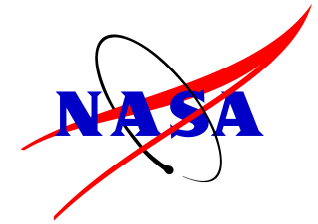
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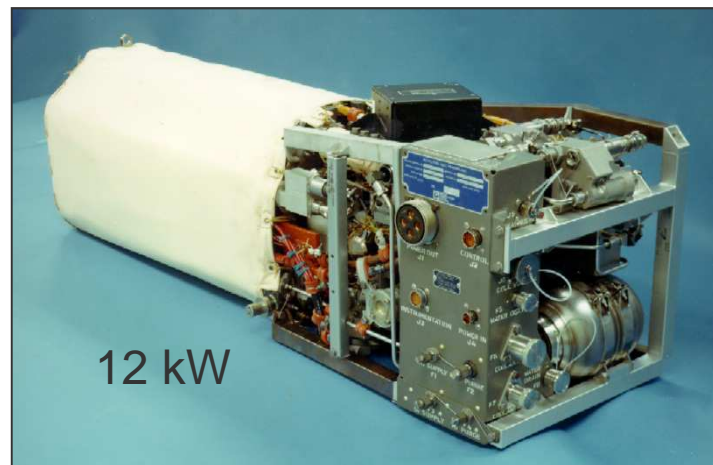
- 1932** Francis Thomas Bacon (UK)  
chemical engineer  
development of practical fuel cells  
in 1959: 5 kW alkaline fuel cell, efficiency 60%



## History of fuel cells



**1960s** NASA modifies Bacon fuel cell  
used to supply on-board drinking water & electricity  
in Apollo missions  
later also in Space Shuttle missions



12 kW  
100 kg  
0.15 m<sup>3</sup>

Source: NASA

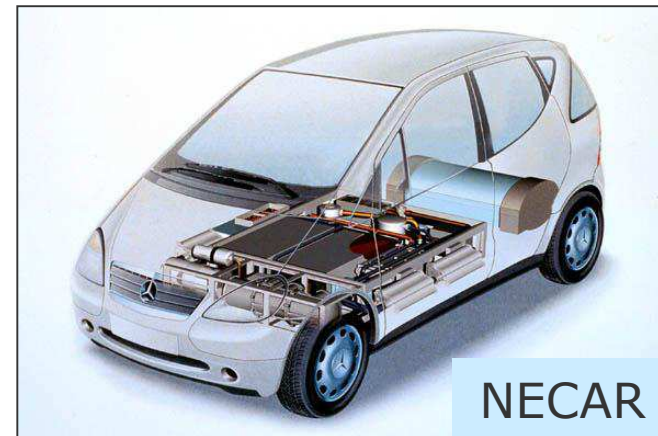
## History of fuel cells

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### 1980s Fuel cells for cars

significant increase in power density over time

all major car manufacturers presented fuel cell prototypes by 2010





## History of fuel cells

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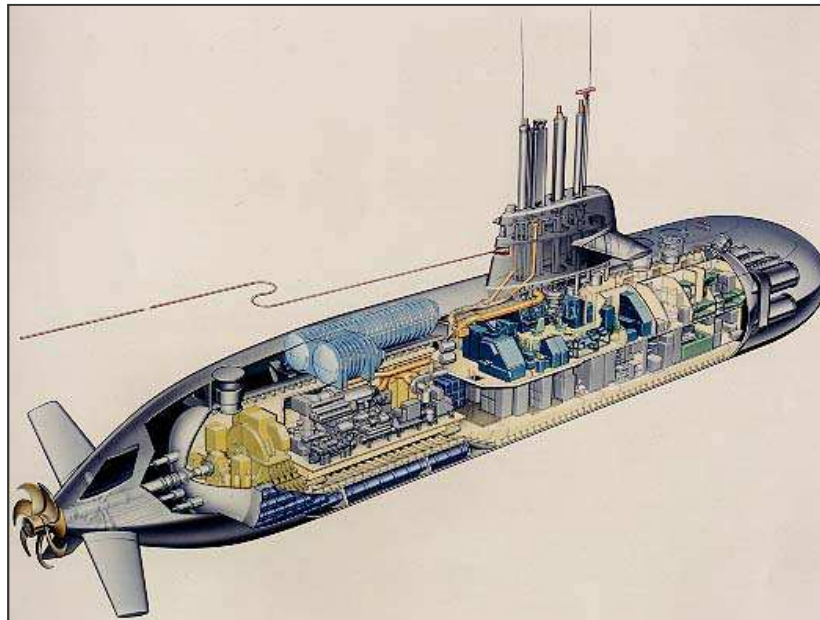
- 2006** Hamburg tests fuel cell bus  
Mercedes Benz Citaro Hybrid-Bus  
9 buses used in public transport



## History of fuel cells

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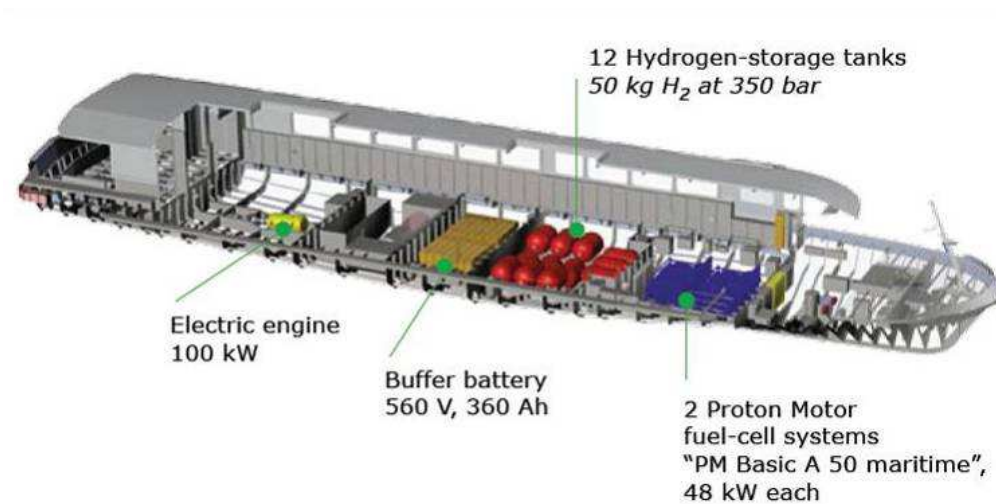
- 2000** U 212 and U 214 submarines (Germany)  
uses PEM fuel cells for air independent propulsion



PEM = Proton Exchange Membrane

## Fuel cells for tourist boats

**2008** Fuel Cell Ship (FCS) "Alsterwasser"  
operated as zero emission ship on lake Alster in Hamburg  
driven by fuel cells  
co-developed by Germanischer Lloyd

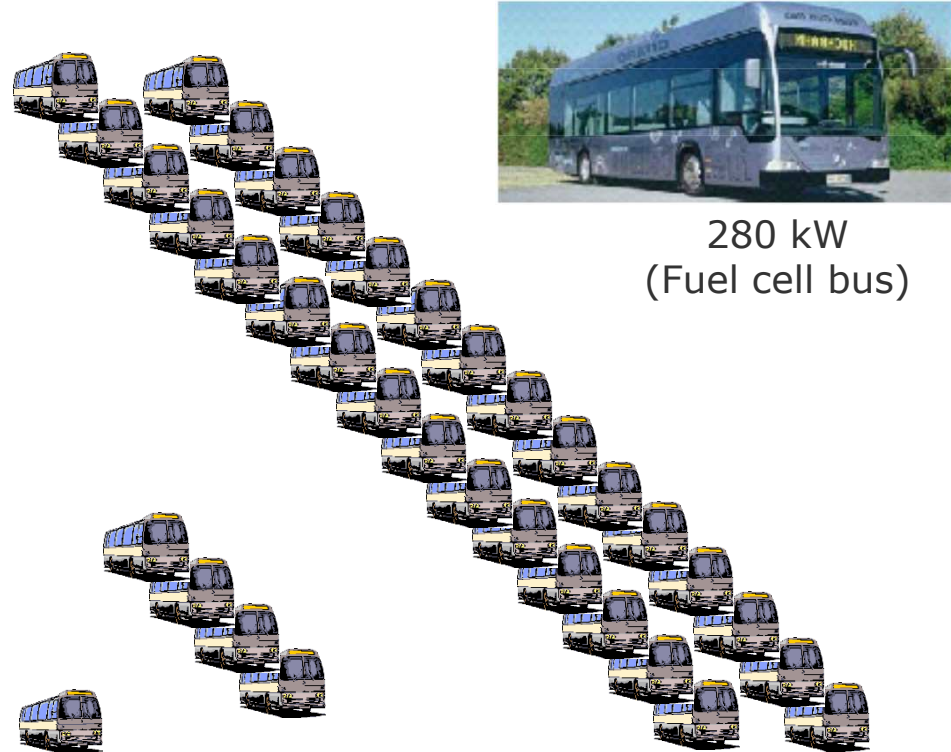


*The propulsion system (Source: Schiffstechnik Buchloh)*

## Fuel cells for ships – A different scale!

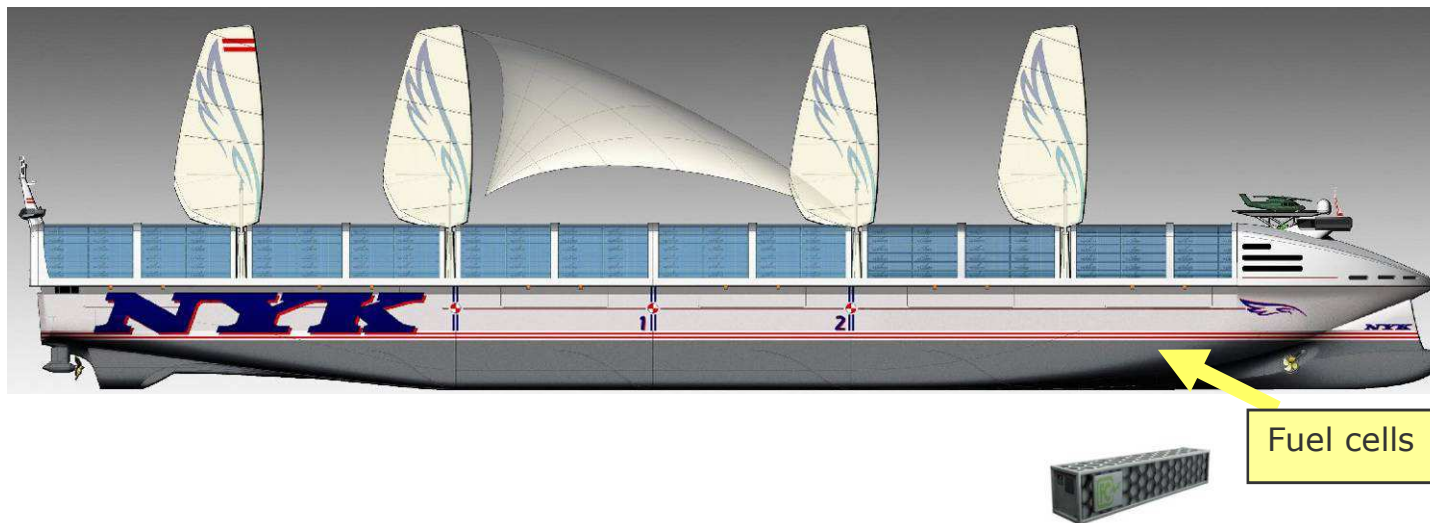


Main Engine        8400 kW  
Auxiliary Engine   1000 kW  
Emerg. Generator  
232 kW



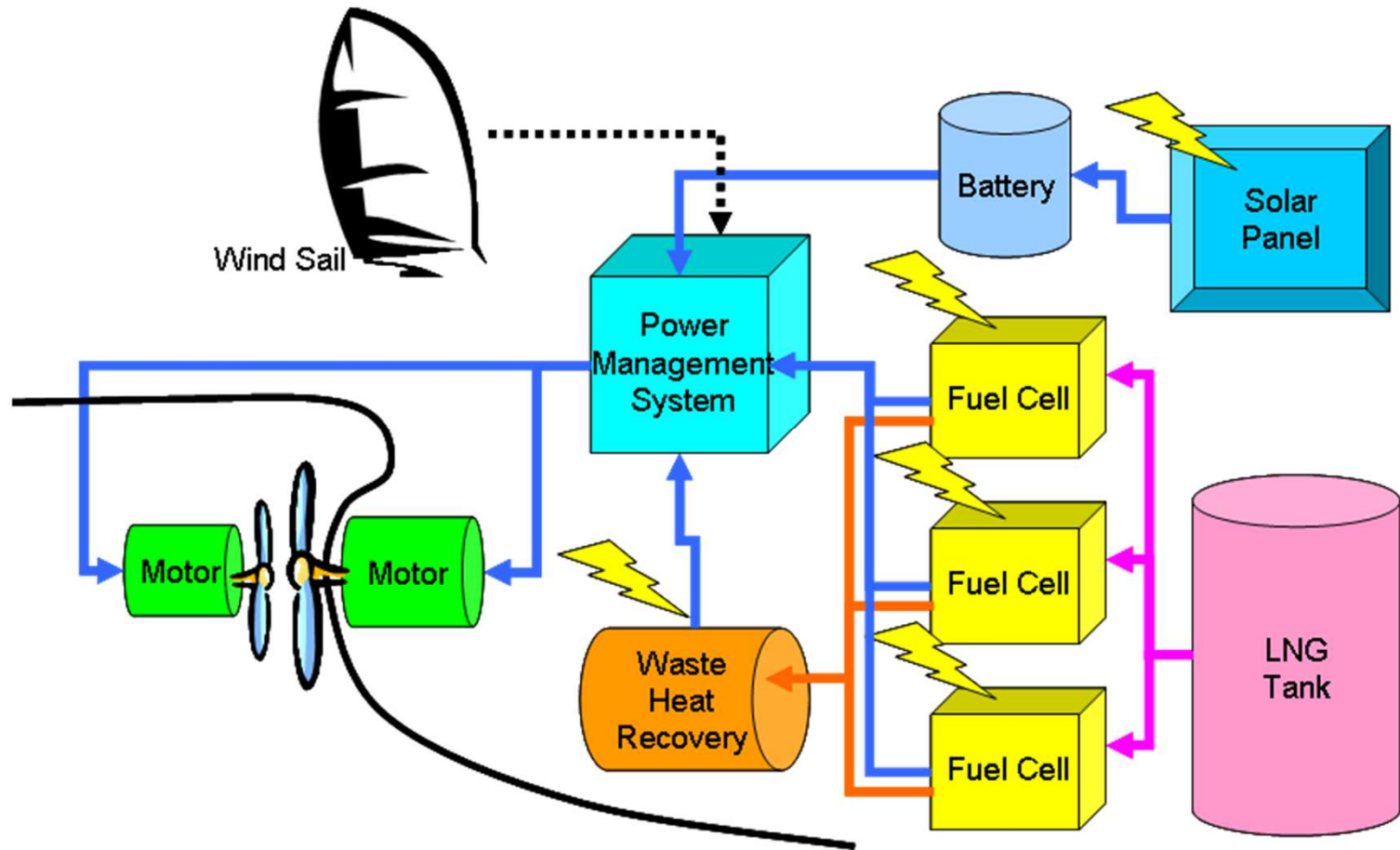
## Future technology

- 2030** Super ECO 2030  
Concept study for large container vessel (NYK)  
various technologies incl. fuel cells  
(claiming 32% less CO<sub>2</sub> due to fuel cells alone)



Source: NYK

# Mix of technologies for power supply



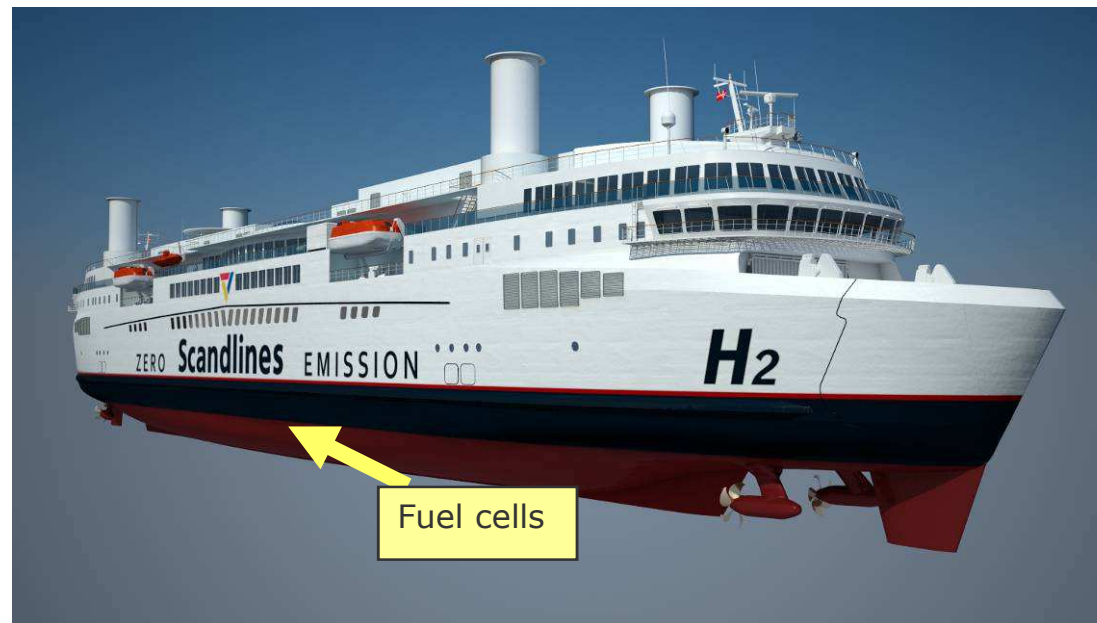
Source: NYK

## Future technology

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**2018 (?)**

Zero-Emission Scandlines project (FutureShip design)  
hydrogen-powered fuel cells  
use excess wind power to generate hydrogen



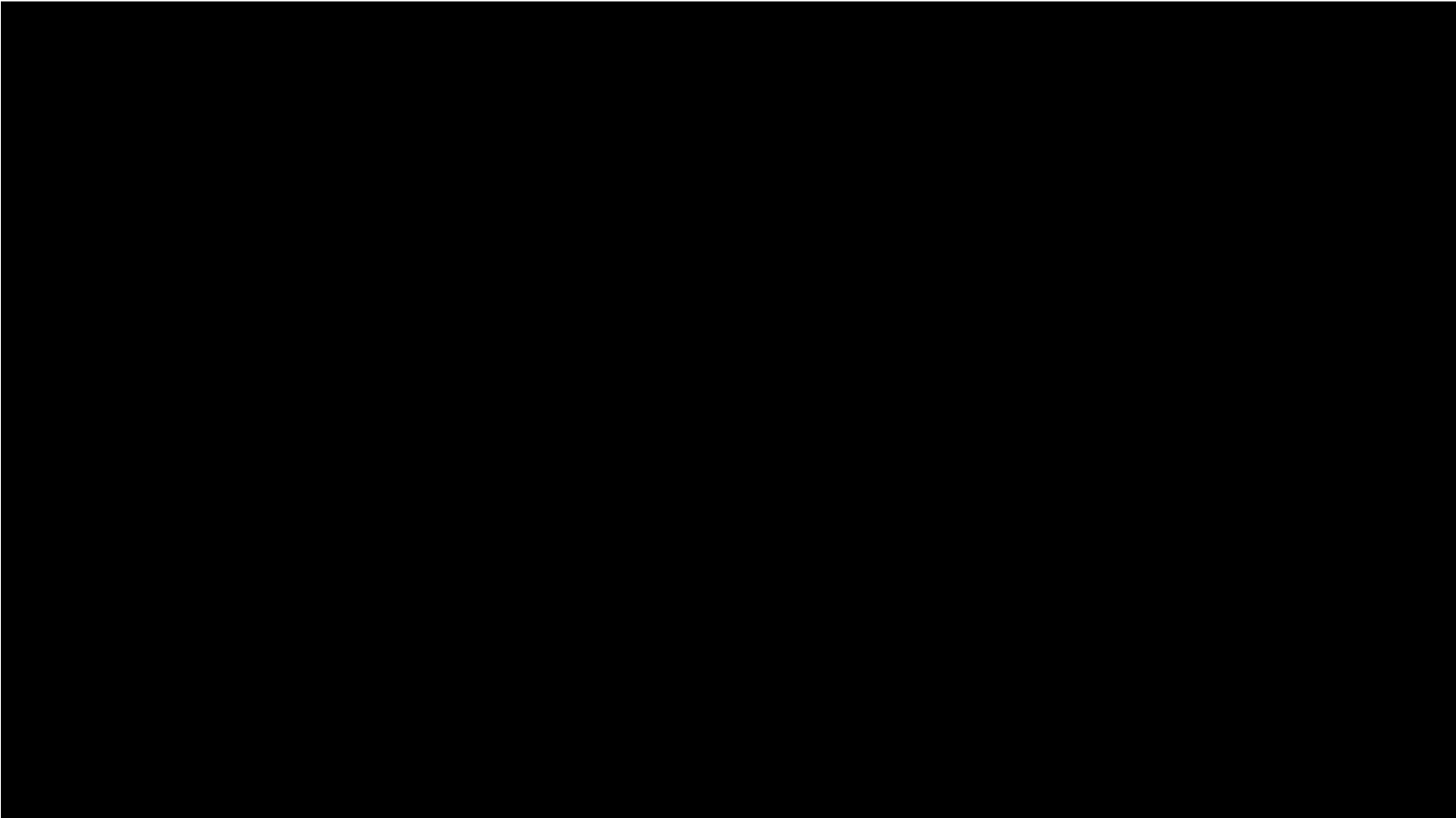
## Future technology

- 2030** GL ZERO (zero-emission Feeder)  
Concept study of Germanischer Lloyd  
fuel cells + batteries (technology of 2010)





# Movie time – Zero-emission feeder



Fuels & Fuel Converters

## Questions so far?

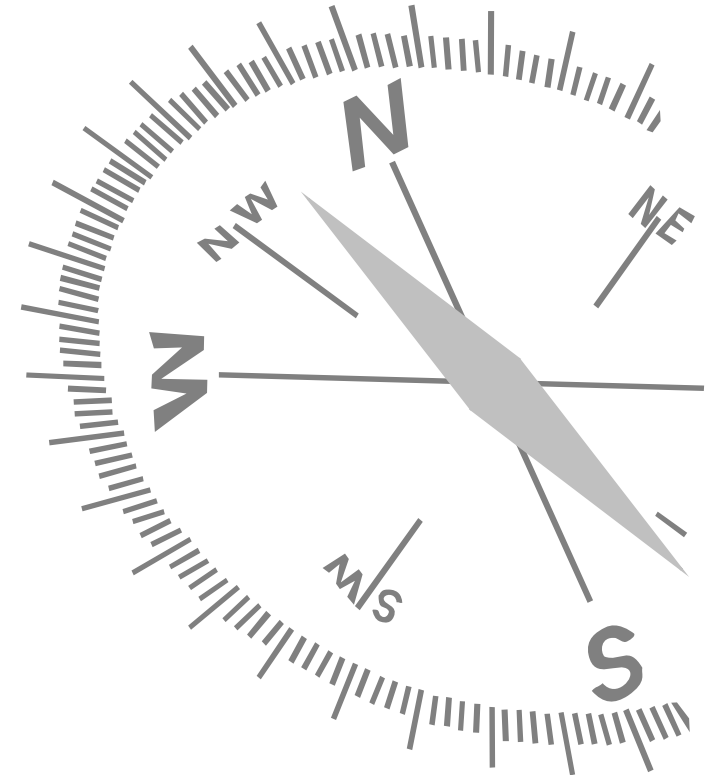
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# Navigator

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- Fuel types
- Natural gas
- Fuel cells
- ➔ Quiz



# Quiz: Do you know your fuel types ?

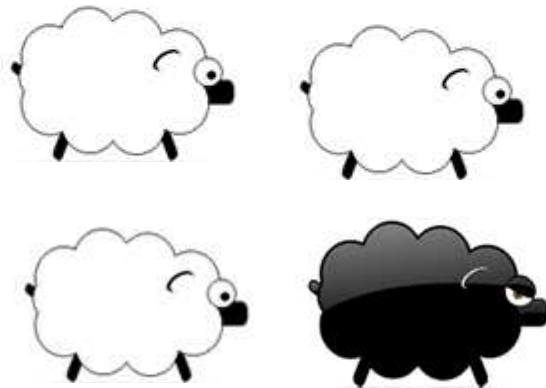
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## What is not true ?

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Longer chains in hydro-carbons lead to...



- a. heavier fuel
- b. lower temperature to evaporate
- c. higher viscosity
- d. lower calorific value

## What is not a fossil fuel ?

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- a. hydrogen
- b. coal
- c. heavy fuel oil
- d. LNG

## Which fuel has the highest calorific value ?

---



- a. alcohol
- b. butane gas
- c. coal
- d. glucose sugar

## Which fuel has the highest calorific value ?

---



- a. Heavy Fuel Oil (HFO)
- b. Marine Gas Oil (MGO)
- c. Marine Diesel Oil (MDO)
- d. Liquefied Natural Gas (LNG)



## Which fuel contains the most gasoil percentage?

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- a. HFO
- b. IFO
- c. MDO
- d. MFO

## IFO 180 is closest in viscosity to...

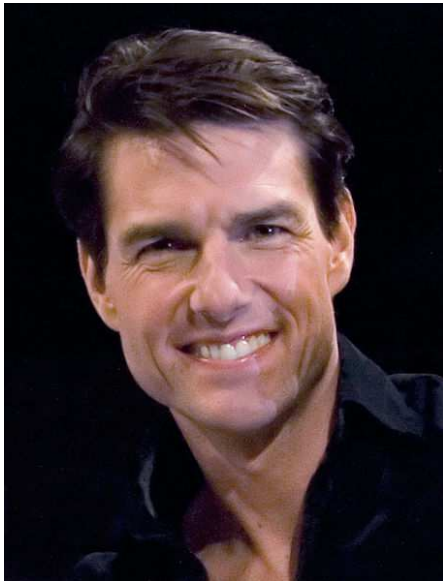
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- a. Milk
- b. Tomato juice
- c. Honey
- d. Tooth paste

## What is the main component in natural gas ?

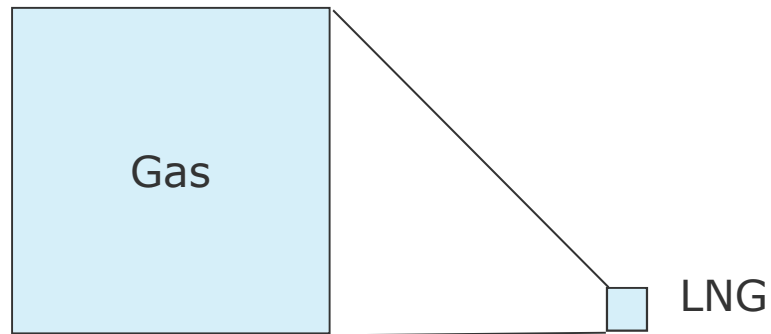
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- a. Ethane
- b. Methane
- c. Nitrogen
- d. Propane

## The volume ratio between LNG and natural gas is...

---



- a. 1:600
- b. 1:200
- c. 1:64
- d. 1:8

## LNG vs HFO

---

For the storage of 1 t of LNG, you need...



- a. roughly the same...
- b. roughly twice the ...
- c. roughly five times the ...
- d. roughly ten times the ...

...volume as for 1 t of HFO

## LNG vs HFO

---

For 1 m<sup>3</sup> of LNG, you get



- a. roughly 10% less than the ...
- b. roughly the same ...
- c. roughly 10% more than the ...
- d. roughly 20% more than the ...

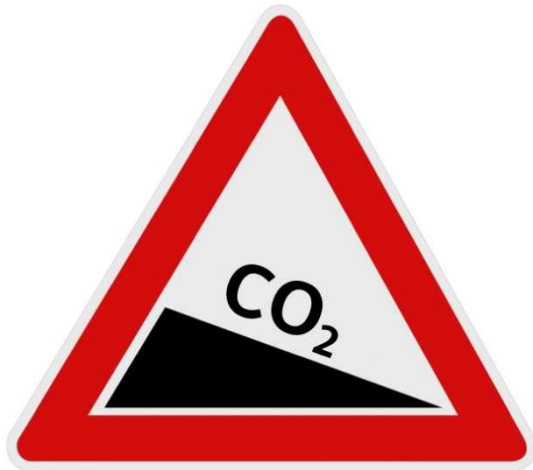
... heat for 1 m<sup>3</sup> of HFO

## LNG vs HFO

---

Compared to HFO, LNG decreases CO<sub>2</sub> emissions by roughly...

- a. 10% ...
- b. 25% ...
- c. 40% ...
- d. 90% ...



... for the same work [g CO<sub>2</sub>/MJ]

## Which fuel has the lowest hydrogen content ?

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- a. wood
- b. coal
- c. oil (HFO)
- d. natural gas



## What is NOT used as fuel for fuel cells ?

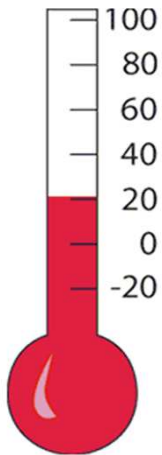
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- a. LNG
- b. liquid hydrogen
- c. liquid oxygen
- d. methanol

## Low-temperature fuel cells ...

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- a. ... have slow start-up
- b. ... have high efficiency
- c. ... do not use LNG as fuel
- d. ... are bulky devices

## High-temperature fuel cells ...

---



- a. ... have slow start-up
- b. ... have low efficiency
- c. ... do not use LNG as fuel
- d. ... are used on portable devices

## Fuel cells generate ...

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- a. ... drinking water & electricity
- b. ... heat & vibrations
- c. ... electricity & vibrations
- d. ... phosphoric acid & heat

# Coffee break

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